



Mayoral Proclamation • City of Madison, Wisconsin



WHEREAS, for 30 years our community has celebrated the Holiday Season with a special gift from UW-Madison chemistry Professor Bassam Z. Shakhashiri. His annual show "Once Upon A Christmas Cheery In The Lab of Shakhashiri" is eagerly awaited by kids and adults alike.

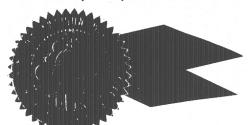
WHEREAS, the show is Dr. Shakhashiri's way of proving that "science is fun" (the legend on a T-shirt he dons for the show). By demonstrating how much fun it can be, Dr. Shakhashiri seeks to impart the joy of discovery that has aroused young minds throughout history. This excitement, he believes, will lure future generations to careers as researchers, entrepreneurs and teachers on whom the nation's continuing economic health and national security will depend. More importantly, he advocates the achievement of literacy in science, mathematics, and technology among those who choose other pursuits. He believes it is essential for the well being of our society that all citizens develop an understanding and an appreciation of science, the benefits of technology, and the potential risks associated with advances in both.

WHEREAS, we are fortunate and thankful that Professor Shakhashiri invites us to his lab to share in the joy of science and to learn how to do scientific experiments safely. His exciting program has become a tradition no only in Madison but is enjoyed by large television audiences on PBS. His guests have included Santa Claus, Bucky Badger, the Mayor of Madison, Nobel prize winners, and the vice president of the United States.

NOW, THEREFORE, BE IT RESOLVED THAT on this 30th Anniversary of "Once Upon A Christmas Cheery In The Lab of Shakhashiri" I, Susan J.M. Bauman, Mayor of the City of Madison, proclaim today, Sunday, December 12, 1999 to be

BASSAM Z. SHAKHASHIRI DAY

FURTHER BE IT RESOLVED THAT on behalf of all Madisonians I salute Professor Shakhashiri for his outstanding accomplishments in enriching the traditions of our Community. Thank you Professor Shakhashiri and Happy Anniversary.



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Signed and sealed this 12th day of December, 1999, at City Hall.





BASSAM Z. SHAKHASHIRI

"Scientist by training, teacher and public servant by trade, advocate by conviction, optimist by nature"—that is the way Bassam Z. Shakhashiri describes himself. As Professor of Chemistry at the University of Wisconsin-Madison, Dr. Shakhashiri finds outlet for all four attributes, to which he might add a fifth: entertainer by avocation.

Dr. Shakhashiri, as a matter of fact, is probably best known to the public at large for his annual program, "Once Upon a Christmas Cheery/In the Lab of Shakhashiri,"; seen on television throughout the country. The science oriented "magic" show has played to packed houses at such varied places as the University of Wisconsin-Madison, the National Academy of Sciences and the Smithsonian's National Air and Space Museum in Washington, and Boston's Museum of Science. The one-hour show as well as two half-hour shows are featured year round on PBS and on other stations. The Christmas Lecture, which is in the tradition of the great British scientist Michael Faraday, is only one demonstration of Dr. Shakhashiri's attachment to hands-on science. He is well known nationally for his development and use of demonstrations in the teaching of chemistry in lecture rooms and laboratories as well as in such less formal settings as convention centers, shopping malls and retirement homes.

He is a guest on TV and radio talk shows across the country and is a regular guest on the Larry Meiller Show on the Ideas Network of Wisconsin Public Radio. He has been featured in newspaper, magazines, national and local radio and television including the *New York Times*, the *Washington Post, Newsweek, Time*, NBC Nightly News, CNN, and the Larry King Show.

A native of Lebanon, Dr. Shakhashiri is the son of a physician who is retired from the U.S. National Institutes of Health in Bethesda, MD. The Shakhashiris, father, mother, son and two daughters, came to the United States in 1957 when Bassam was 18 years old with one year of college (at the American University of Beirut) behind him. He completed undergraduate work at Boston University (Class of '60) with an A.B. degree in chemistry, served as a teaching fellow at Bowdoin College for one academic year and then earned master's and Ph.D. degrees in chemistry at the University of Maryland ('64 and '68 respectively).

After a year of post-doctoral research and two years as a junior member of the chemistry faculty at the University of Illinois, Urbana, Dr. Shakhashiri joined the faculty of the University of Wisconsin in 1970, a position he has held since. In 1977 he was the founding chair of the University of Wisconsin System Undergraduate Teaching Improvement Council. In 1983 Dr. Shakhashiri founded the Institute for Chemical Education (ICE) and served as its first director.

His SCIENCE IS FUN! presentations in Madison at shopping malls, the Capitol Square, the Great Walk of the Muscular Dystrophy Association, KIDS EXPO, etc., as well as his visits to schools and colleges, have reached tens of thousands of students, their teachers, and parents throughout Wisconsin.

Dr. Shakhashiri has given about 1000 invited lectures and presentations in the United States and other countries. He has co-authored several publications including Manual for Laboratory Investigations in General Chemistry; Workbook for General Chemistry Audio-Tape Lessons; Chemical Demonstrations: A Handbook for Teachers of Chemistry, Volumes 1, 2, 3 and 4; and semi-programmed booklets on equilibrium, kinetics, and organic chemistry. The Shakhashiri Chemical Demonstrations Videotapes and Videodiscs were published in 1991 by Holt, Rinehart and Winston, Inc. and Saunders College Publishing Company. His publications, television programs, and web site (www.scifun.chem.wisc.edu) are the bases for what thousand of chemists and teachers present annually during National Chemistry Week, National Science and Technology Week, and on a daily basis in classrooms and science museums across the country and elsewhere. Another of his pioneering efforts is an interactive chemistry exhibit on permanent display since 1983 at the Chicago Museum of Science and Industry. On June 26, 1984 he was sworn-in as Assistant Director of the National Science Foundation for Science and Engineering Education by the President's Science Adviser. In this position he was the principal education officer of the federal agency chiefly concerned with research in the natural sciences and engineering. As such, he was responsible for the design and administration of a wide variety of programs to improve all levels of education in mathematics, engineering and the sciences.

He presided over the rebuilding of the NSF efforts in science and engineering education after they had been essentially zeroed-out in the early 1980's. He established NSF's integrity and credibility by the quality of the staff he hired, the design of new programs, and the securing of funds. His visionary strategic plans were designed and implemented with the aid of a most distinguished national advisory committee which he appointed and with the support of Congress. Systemic reform programs were launched;

elementary and middle school programs were introduced along side rejuvenated and expanded high school programs; undergraduate programs were created and increased; and graduate fellowships and traineeships were buttressed. His leadership and effectiveness in developing and implementing national programs in science and engineering education became legend and have helped set the annual NSF education budget at over \$600 million. In September of 1990 he returned to Madison and has taught introductory level chemistry to over 600 students annually. He vigorously continues his advocacy for both increasing the flow of talent to careers in science and achieving science literacy by the public at-large. He was elected to a variety of faculty committees including the Executive Committee of the Physical Science Division, the Executive Committee of the Graduate School, and the Honorary Degrees Committee. He is a leader in the Freshman Learning Community Program and the Women in Science and Engineering Program.

Dr. Shakhashiri is a member of many scientific and educational organizations including the American Chemical Society, in which he has held numerous leadership positions at the local and national levels. In 1986, he was elected a fellow of the American Association for the Advancement of Science. In 1987, he was elected an honorary member of the South Carolina Academy of Science and in 1990 an honorary member of the Alabama Academy of Science.

Among his many awards are the 1977 Kiekhofer Distinguished Teaching Award from the University of Wisconsin-Madison, and the 1979 Manufacturing Chemists Association Catalyst Award. He is the youngest recipient of two of the American Chemical Society's most coveted recognitions—the James Flack Norris Award for Outstanding Achievement in the Teaching of Chemistry (1983) and the ACS Award in Chemical Education (1986). In 1982 he was given the Ron Gibbs Award of the Wisconsin Society of Science Teachers. In 1995 he was cited in the *Year Book of Encyclopaedia Britannica* as the "dean" of lecture demonstrators in America. In 1998 he was given the Sacred Heart University Presidential Initiative Award in recognition of his national contributions to advancing science education. He is the recipient of five honorary doctoral degrees.

Dr. Shakhashiri serves on many national boards and governing bodies including the Board for the Merck Institute for Science Education; the Mathematics and Science Initiative Board of Teach for America; the National Advisory Board of *The Scientist*; the national board of the Center for Chemical Education at Miami University (Oxford, OH), the board of the Center for the Advancement of Science Education of the Wisconsin Academy of Sciences, Arts, and Letters; and is affiliated with the National Institute for Science Education at the University of Wisconsin-Madison.



Although the traditional marker for 30th anniversaries is pearls, for a chemistry anniversary, the 30th chemical element may be more appropriate. This element is zinc. Perhaps not as charming as pearls, zinc is certainly more important in our everyday lives.

Zinc is a metallic element, whose atomic number is 30. The element does not occur as the free metal in nature. Instead, it is most commonly found in ores combined either with oxygen, as in calamine, or with sulfur, as in zincblende. Metallic zinc has been extracted from these ores for nearly a thousand years by heating the ores with coal or charcoal. Zinc vapor from this process is condensed to solid zinc metal.

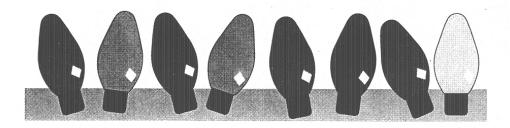
Metallic zinc is one of the more chemically active metals. Its activity is exploited in a number of commercial products. One such application (and the major application) of zinc is in galvanized iron and steel. Because zinc is more chemically active than iron, a coating of zinc on iron materials such as sheet metal protects the iron from corrosion (rusting). The average automobile contains about 17 pounds of zinc for this purpose. Another application that uses the chemical activity of zinc is in dry cells and batteries. These cells contain metallic zinc, among other materials. Zinc supplies the electrons that form the electric current that drives a battery-operated device.

Zinc is combined with other chemical elements to produce many every-day products. One of the oldest of these combinations is that of zinc with copper. This combination produces brass, a hard yellow alloy used for making many decorative items and the musical instruments that comprise the "brass" section of an orchestra. Zinc forms a compound with the element sulfur, zinc sulfide. This compound is a phosphor, that is, it glows when it absorbs energy. It is used to produce the coatings inside fluorescent lamps, the light-emitting surfaces inside television tubes, and many glow-in-the-dark novelty items. Zinc oxide, the compound of zinc and oxygen, has many uses in health products. It is an effective absorber of ultraviolet radiation, and because of this, it is used in one type of sun screen. Zinc oxide also has a soothing effect on the skin and is used in medicated skin powders.

Zinc is an essential trace mineral required by humans and animals. It plays an important role in many biochemical processes. Zinc is essential for gene expression and for the metabolism of nucleic acids. This makes it important in cell growth and differentiation. It is involved in a wide range of enzymes, and plays a structural role in cell membranes and hormone receptors. The U.S. Government Recommended Daily Allowance for zinc is 15 milligrams. Most American adults, however, obtain only about 10 milligrams in their diets. Foods that are particularly rich in zinc include meats, seafood, peanut butter, and cheese. One of the hidden uses of zinc is in the contemporary U.S. cent. Although it may look like copper, this coin, since 1983, has been 98% zinc with a thin coating of copper to disguise it. For more information about the 30th element, see the Web site of the International Zinc Association at www.zincworld.org.



Michael Faraday, the noted English physicist and chemist, lived from 1791 to 1867. He was a gifted lecturer, and he began giving his Christmas Lectures for children at the Royal Instituion of Great Britiain in the 1840s. Faraday loved simplicity, and he had a strong sense of the dramatic. His audience entered wholeheartedly into the world of science with him as guide. His ideas were still considered very unorthodox at that time, and children, who had not vet adopted conventional ideas, would react enthusiastically to the ones he presented. Eventually, the lectures became very popular, and even the Prince of Wales attended and learned about the mysteries of electricity. Faraday sought to awaken the sense of wonder in his listeners. He knew that once a person could be made to wonder about the world, it was only a short step to studying it. He strove to point out that if you looked closely at the most ordinary thing, such as the force of gravity, it ceased to be ordinary and became somehow miraculous. Faraday did all he could to urge his listeners to see and judge for themselves, to experiment-to question nature directly-whenever anyone discovered something out of the ordinary.





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