
Hydrogen to Copernicium: Postage Stamps as Cultural Icons in the IYC

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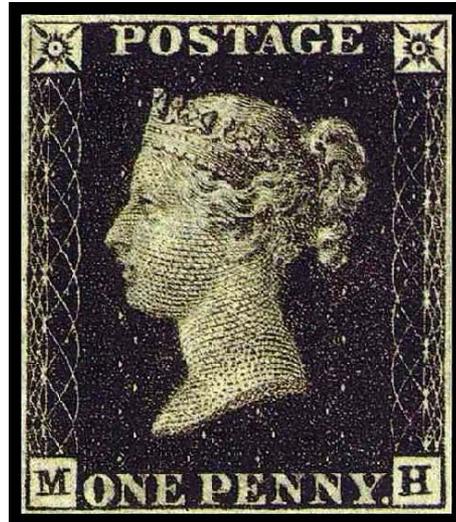
Once upon a time...

- Until the 1830's, letters were paid by the recipient not the sender.



- Sir Rowland Hill, an English teacher, inventor, and social reformer (1795-1879), introduced postage stamps in 1840.

Philately



Penny Black
1840

- From the Greek *phil-* “love” and *ateleia* “exempt from tax”.
- The collecting and study of postage stamps and other postal matter as a hobby or an investment.

and some are issued for a reason...

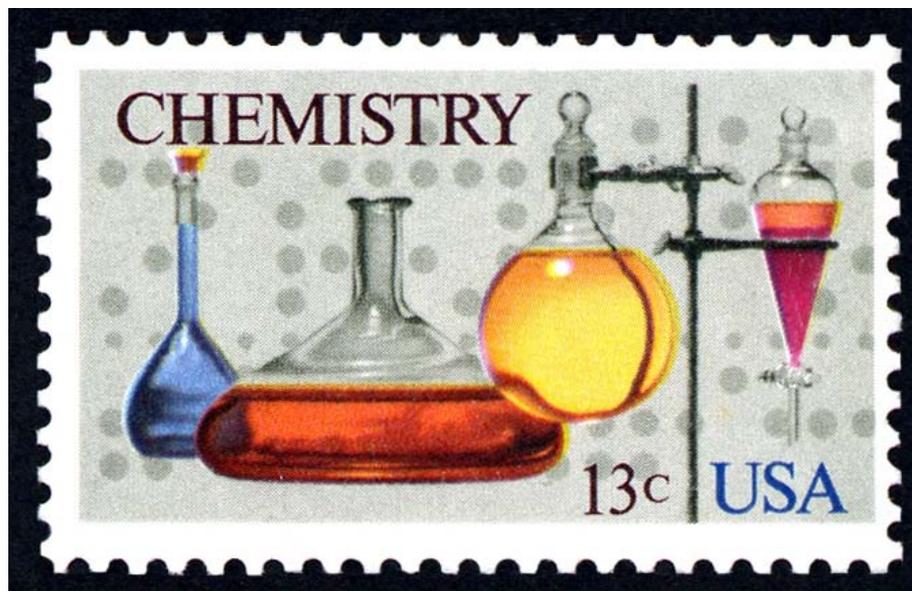
- Many stamps have been issued to commemorate events and educate the public: history, art, literature, *science...*



- Attractive way to illustrate oral presentations, papers, etc.
- Inexpensive and effective teaching tools!

Chemistry on Stamps

- *Chemophilately (chemical philately):* the philatelic study of chemistry (Rappoport, *Z. Acc. Chem. Res.* **1992**, 25, 24-31).



Heilbronner, E.; Miller, F.A. *A Philatelic Ramble through Chemistry*; Wiley-VCH, 1998.

The Tree of Science



Once upon a time...

The Four Classical Elements in Greece



air

fire

water

earth

air

fire



earth

water

air



fire

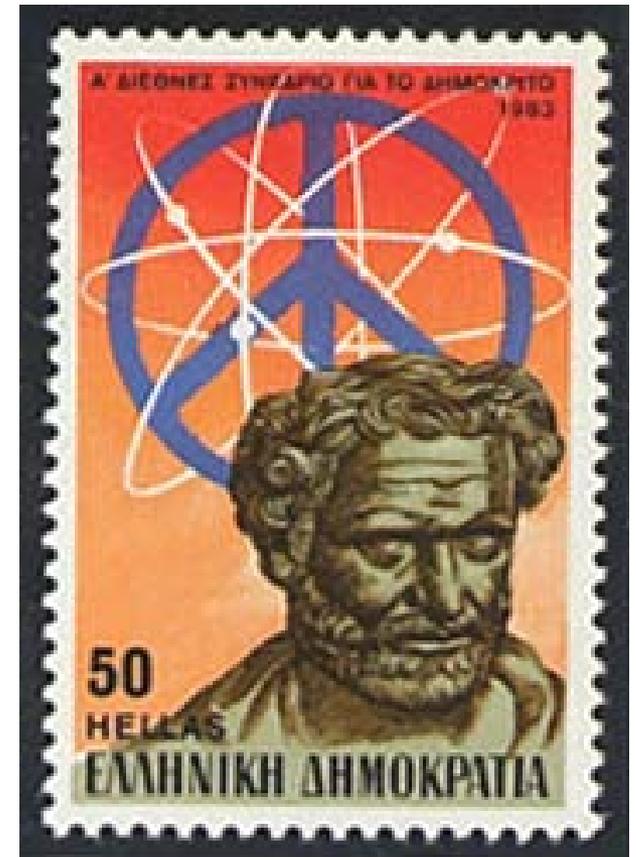
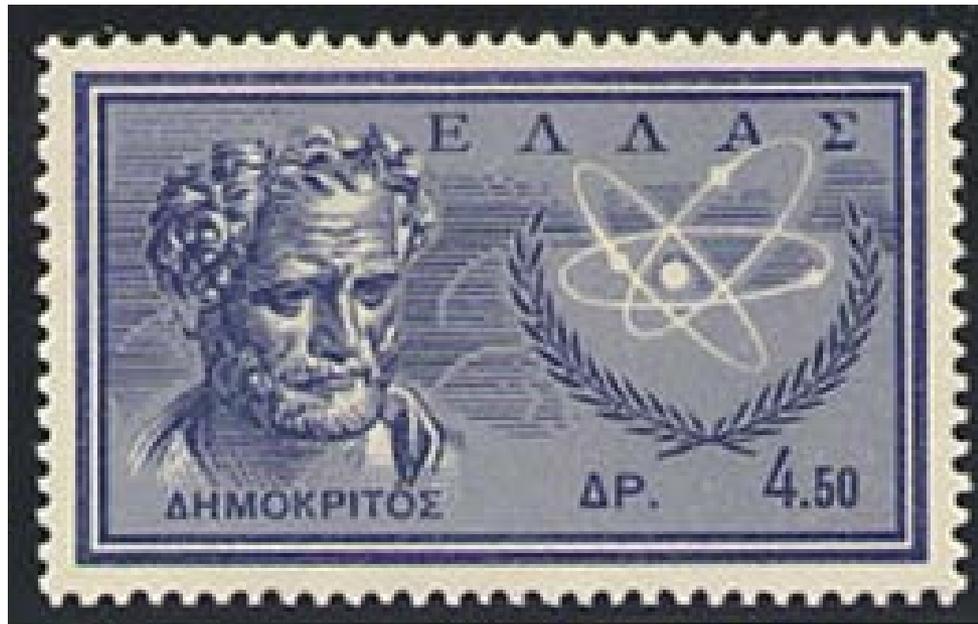


earth

water

Democritus (460-370 BC)

Matter is made up of extremely small, indivisible particles: “atoms”.



Alchemy in the Middle Ages

Avicenna



Avicenna

Rhazes



Paracelsus

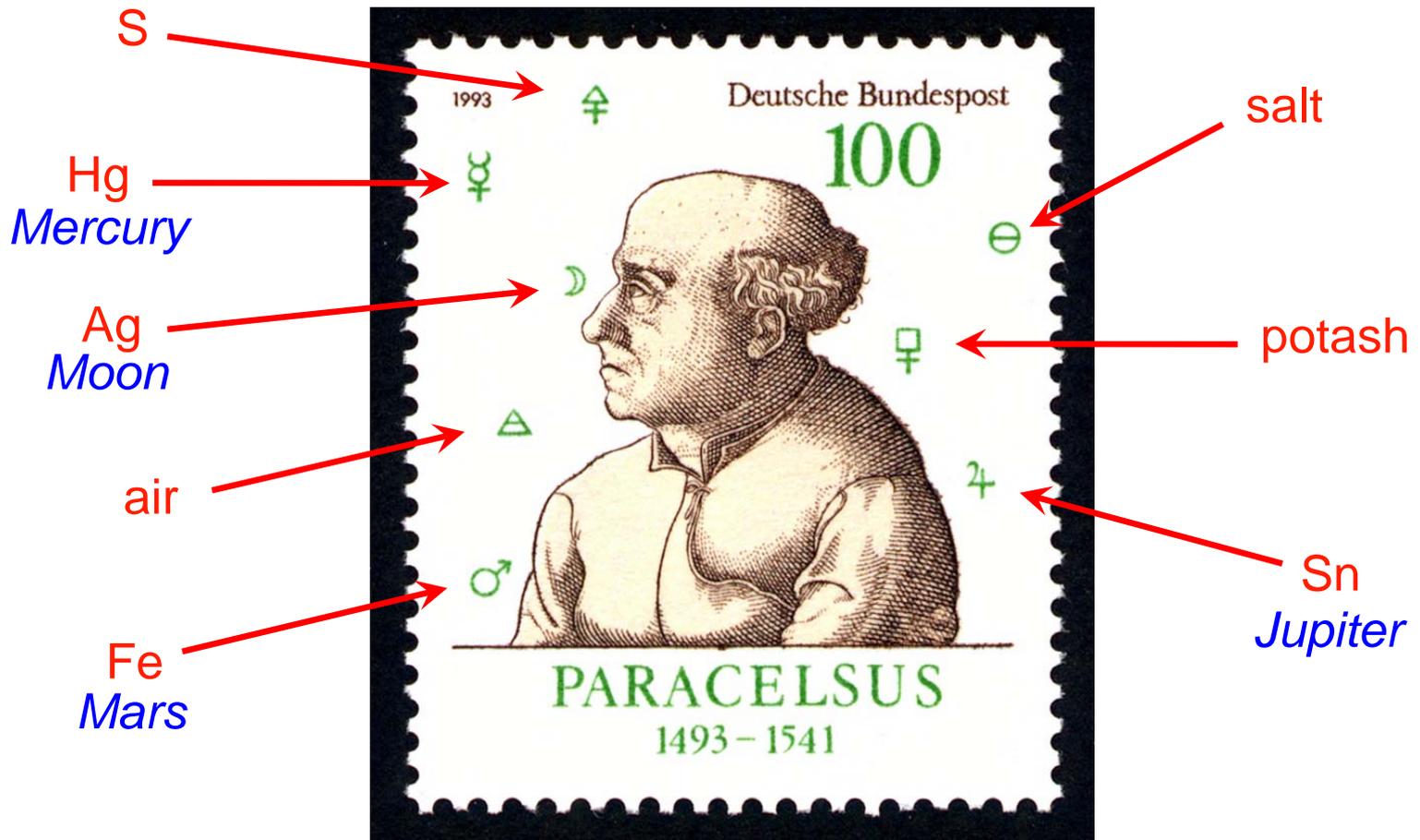


Agricola

Theophrastus Philippus Aureolus Bombastus von Hohenheim

Paracelsus

(1493-1541)



Robert Boyle (1627-1691)



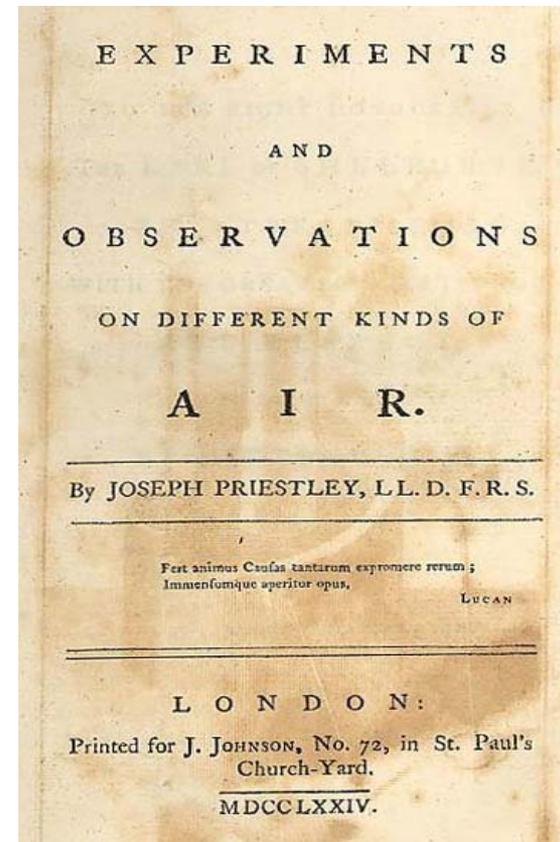
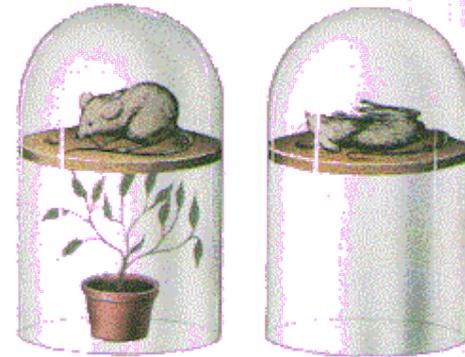
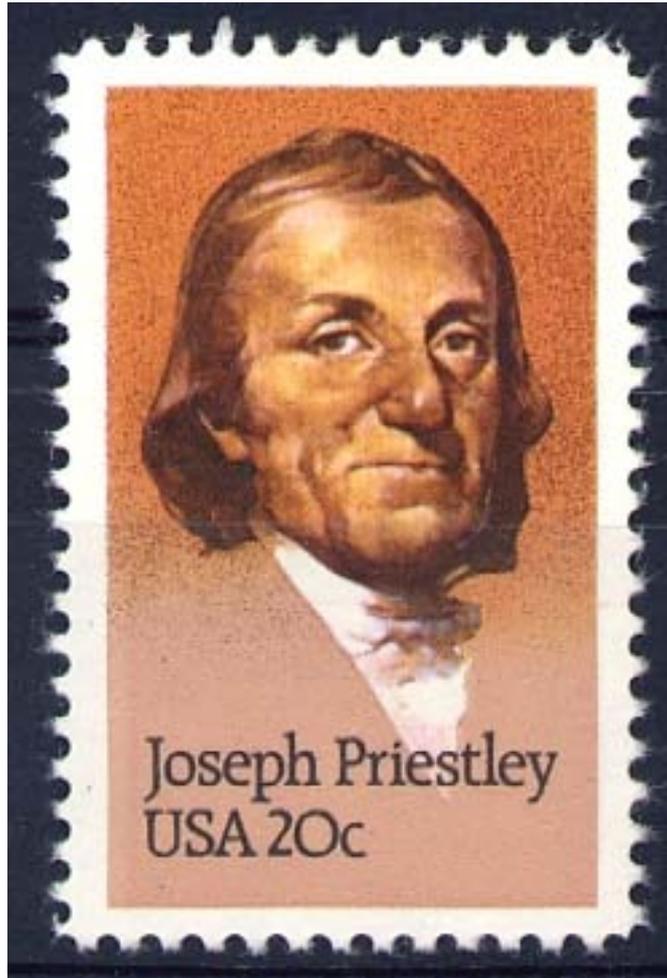
Issued on February 25, 2010: 350th anniversary of the Royal Society.

Antoine Lavoisier (1743-1794)

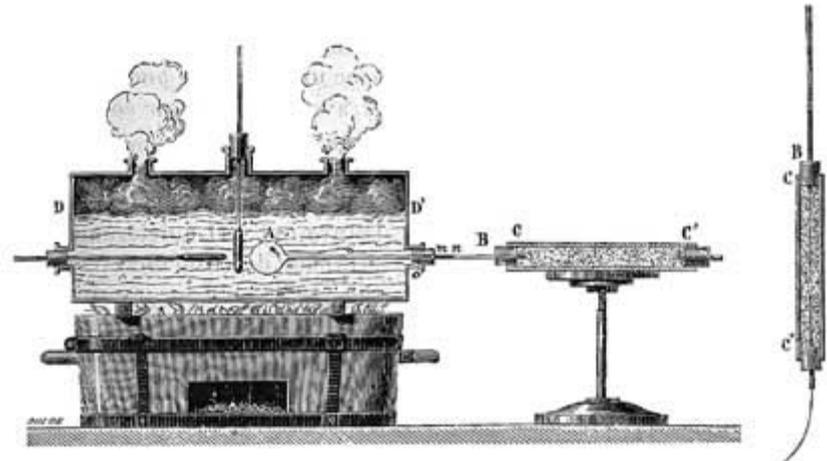
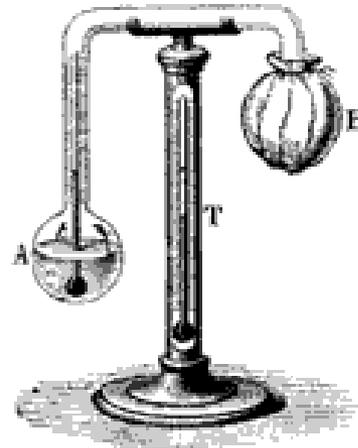


maximum card

Joseph Priestley (1733-1804)



Joseph Louis Gay-Lussac (1778-1850)



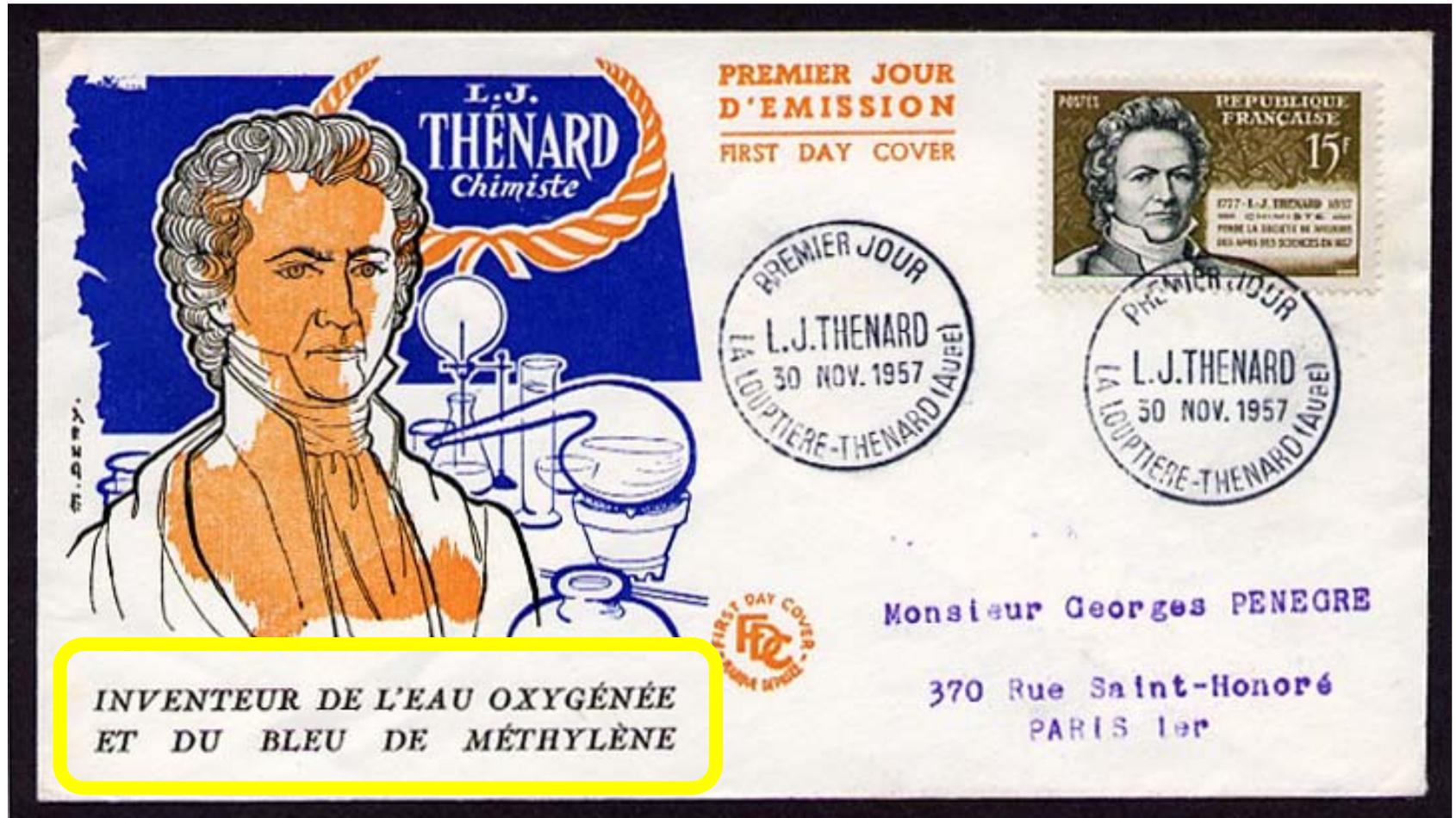
*Lorenzo Romano Amedeo Avogadro di Quaregna e Cerreto
(1776-1856)*



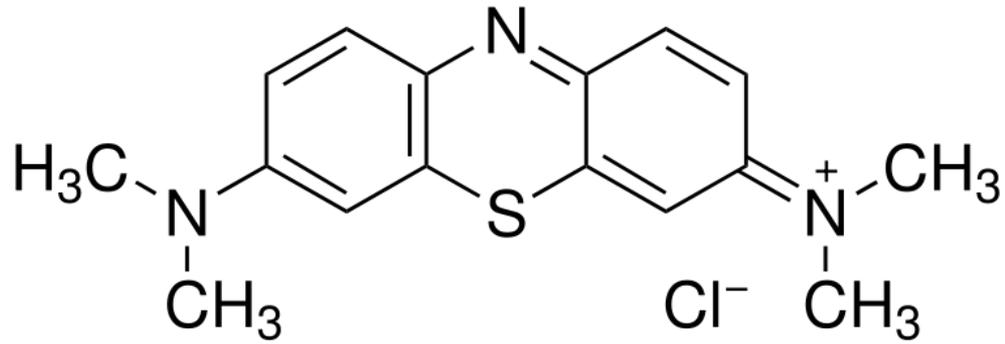
Avogadro's Law: "equal volumes of gases, at the same temperature and pressure, contain the same number of molecules".

$$N_A = 6.022 \times 10^{23}$$

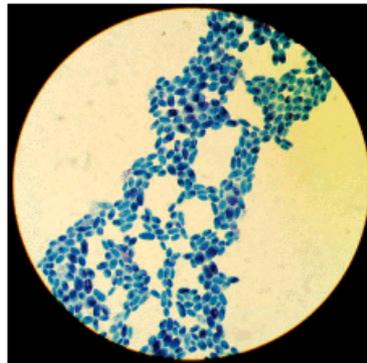
Louis Jacques Thénard (1777-1857)



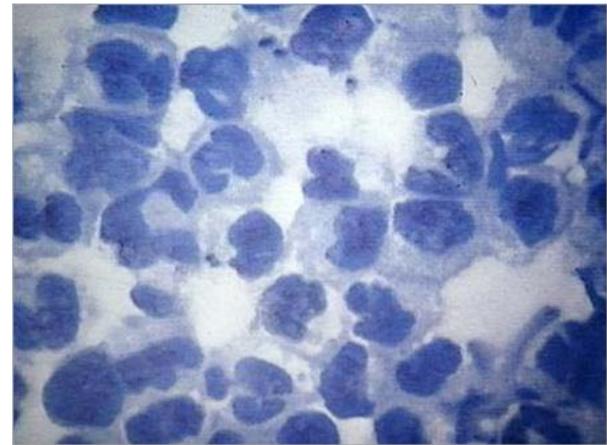
Methylene Blue



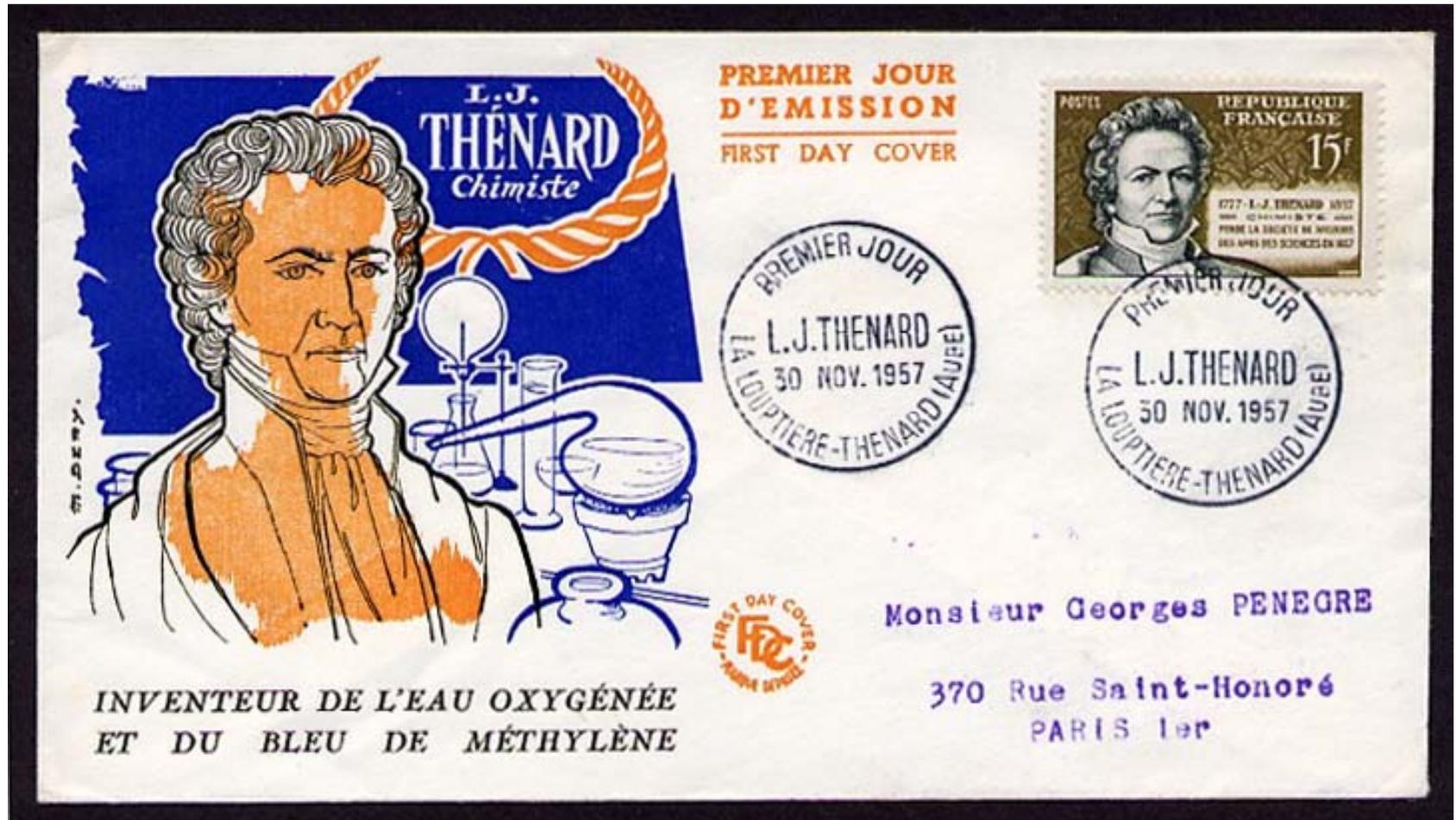
redox indicator



biological stain

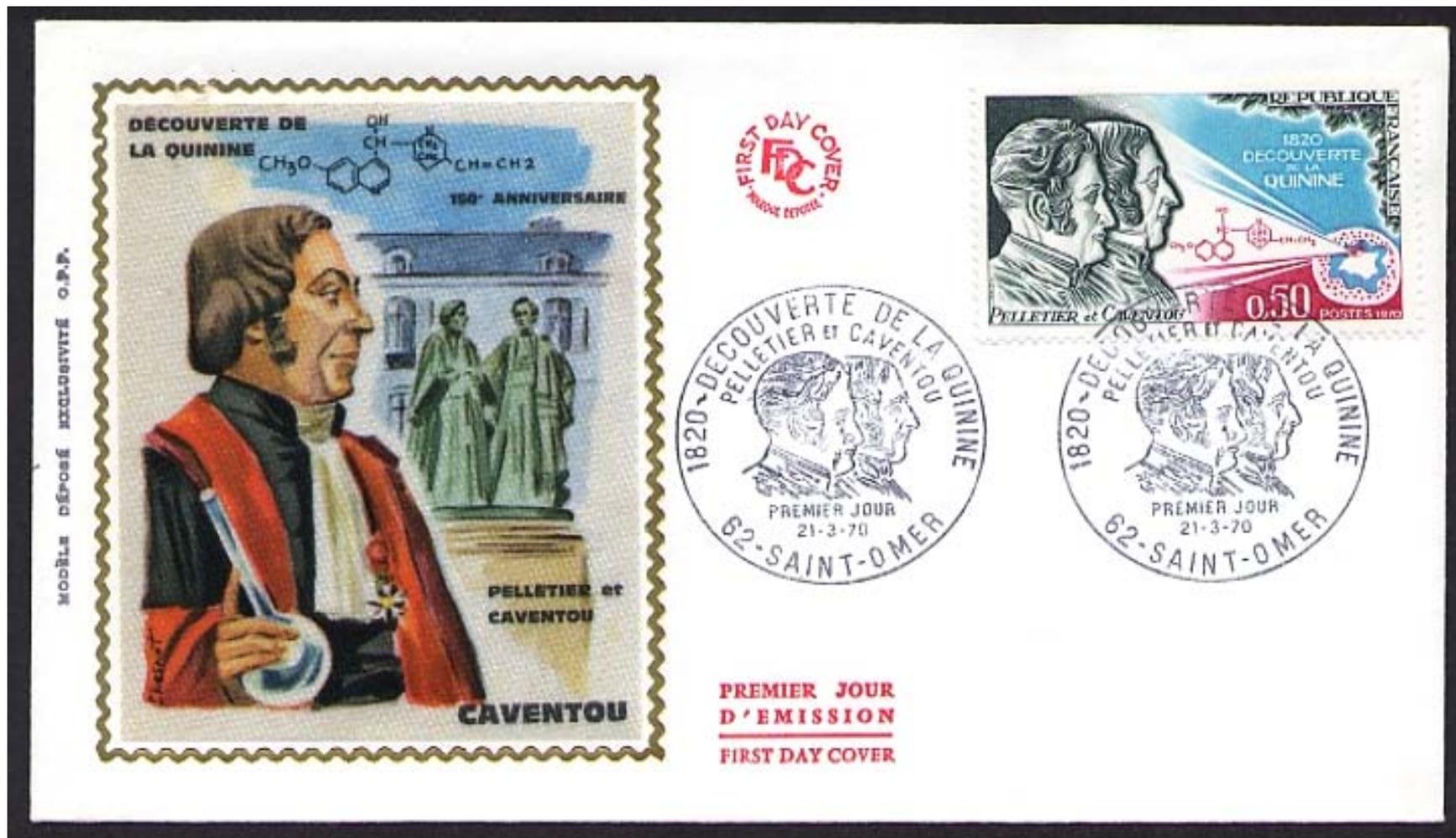


Louis Jacques Thénard (1777-1857)



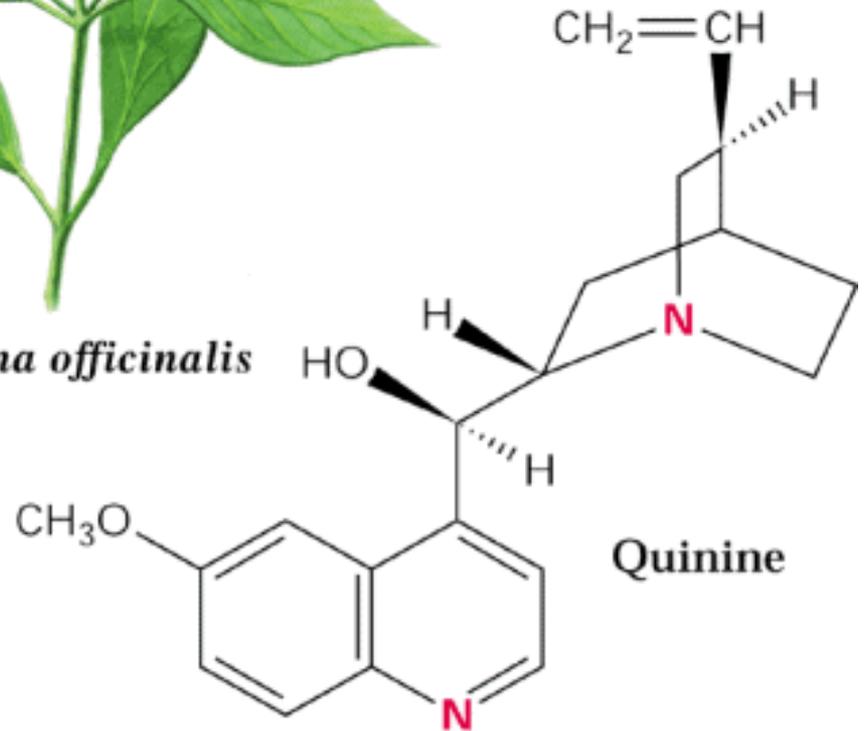
Cobalt blue (Thénard's blue): $\text{CoAl}_2\text{O}_4 \dots!$

Pelletier & Caventou: the discovery of quinine (1820)

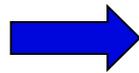




Cinchona officinalis



Quinine



Justus von Liebig (1803-1873)



Major contributions to agricultural, biological and organic chemistry.

Justus von Liebig (1803-1873)

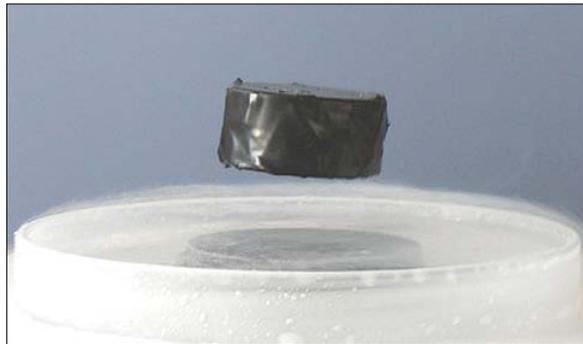


Franz-Joseph Müller von Reichenstein (1740-1825)



- Hungarian mineralogist who discovered tellurium (Te) in 1782.

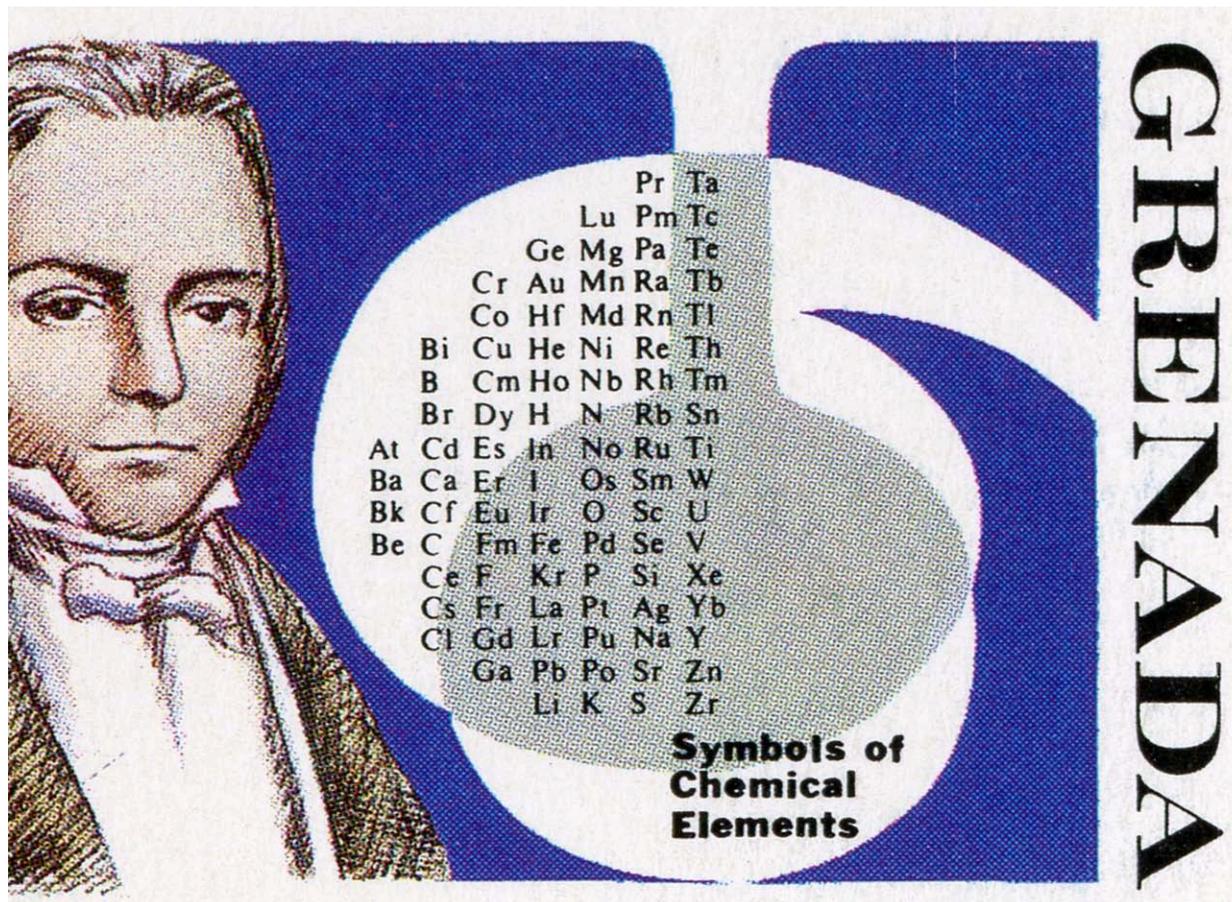
Johan Gadolin
(1760-1852)



- Founder of Finnish chemistry & discoverer of yttrium (Y) ca. 1794

Jöns Jacob Berzelius (1779-1848)

- Developed a system of symbols for the chemical elements.



Jöns Jacob Berzelius (1779-1848)



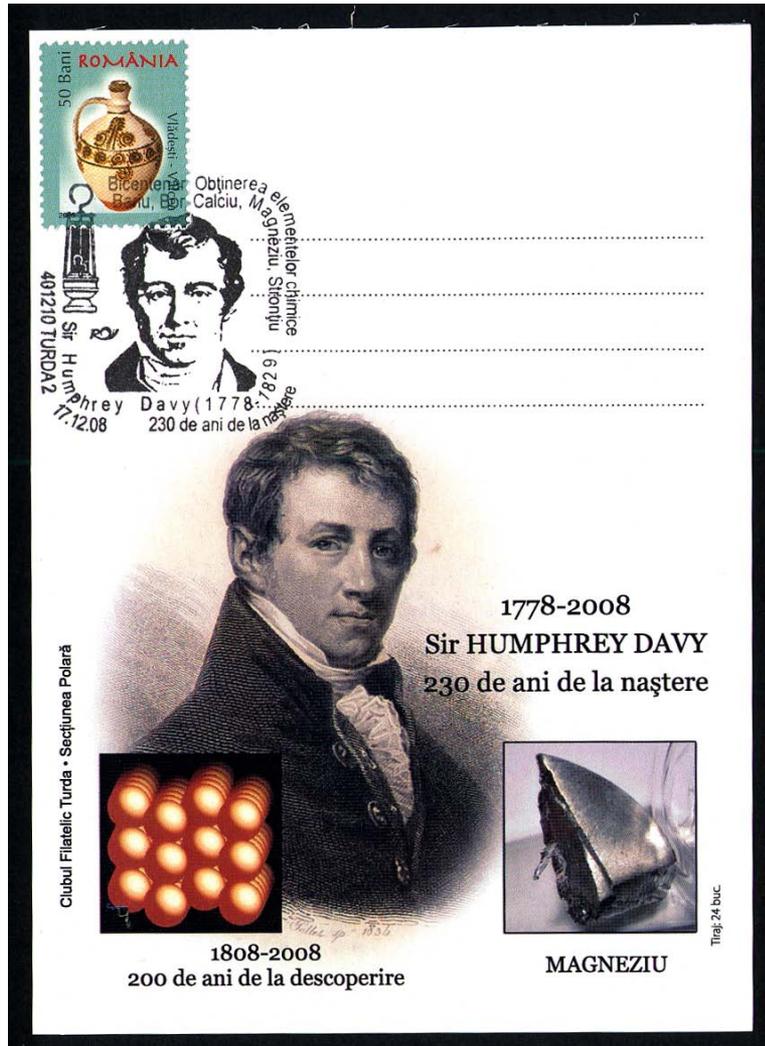
- Coined the terms “catalysis”, “polymer”, “isomer”, and “allotrope”.

Bicentennial of the Karolinska Institute (1810-2010)

- Berzelius also discovered silicon, selenium, thorium, and cerium.



Humphry Davy (1778-1829)



- Discovered Na, K, Mg, Ca & Ba
- Developed miner's lamp



Johann Wolfgang Döbereiner (1780-1849)

Triads, triads everywhere...



Leopold Gmelin (1788-1853)



Gmelin's "Periodic Table" (1843)

		O		N		H			
F	Cl	Br	I			Li	Na	K	
	S	Se	Te			Mg	Ca	Sr	Ba
	P	As	Sb			Be	Ce	La	
	C	B	Bi			Zr	Th	Al	
		Ti	Ta	W		Sn	Cd	Zn	
			Mo	V	Cr	U	Mn	Ni	Fe
				Bi	Pb	Ag	Hg	Cu	
			Os	Ir	Rh	Pt	Pd	Au	

Dmitri Mendeleev (1834-1907) and the Periodic Table

СТОЛЕТИЕ ПЕРИОДИЧЕСКОГО ЗАКОНА Д.И.МЕНДЕЛЕЕВ

ПЕРВЫЙ ДЕНЬ PREMIER JOUR

ПОЧТАМТ 20.VI.1969 Д.И.МЕНДЕЛЕЕВ

СССР МОСКВА PREMIER JOUR 1969 ПОЧТА СССР 30

СТОЛЕТИЕ ПЕРИОДИЧЕСКОГО ЗАКОНА Д.И.МЕНДЕЛЕЕВ

*Орбита ~~Солнца~~ ~~Земли~~ ~~и~~ ~~Луны~~,
основанная на ~~гравитационной~~ ~~теории~~ ~~Ньютона~~ и ~~Гюкенса~~,
Д. Менделеева.*

Li = 50	Zn = 90	P = 180				
V = 51	Ni = 94	Sn = 182				
Ce = 52	Ko = 96	W = 186				
Mg = 55	Rh = 104	Pt = 197,4				
Sc = 56	Ra = 109	Ir = 196				
Ni = Co = 59	Pt = 105,6	Os = 197				
H = 1	Be = 8	B = 22	Cu = 63,4	Mg = 101	Mn = 200	
He	Be = 9,4	Mg = 34	Zn = 65,2	Co = 82	Fe	
He	Al = 11	Al = 22,4	P = 68	Ni = 116	As = 175?	
				70	Su = 118	
				75	Si = 128	Bi = 210?
				29,4	Fe = 123?	
				30	P = 127	
				85,4	Ca = 153	Fl = 204
				176	As = 132	Pt = 207?
				12		
				74		
				5		
				18		
				17		
				18		
				17		

18 $\frac{II}{17}$ 69.

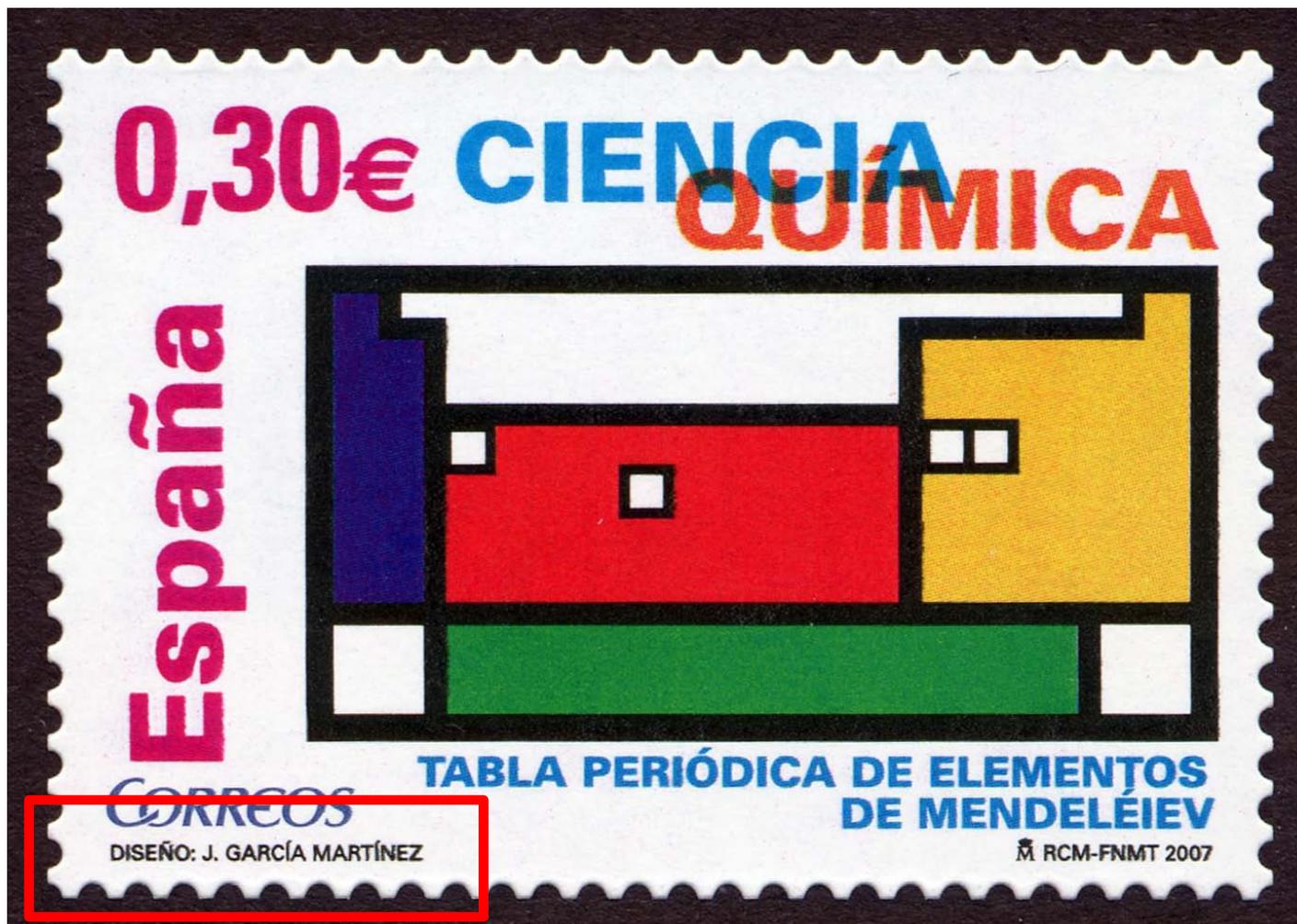
1869 Periodic Table

Ueber die Beziehungen der Eigenschaften zu den Atomgewichten der Elemente. Von D. Mendelejeff. — Ordnet man Elemente nach zunehmenden Atomgewichten in verticale Reihen so, dass die Horizontalreihen analoge Elemente enthalten, wieder nach zunehmendem Atomgewicht geordnet, so erhält man folgende Zusammenstellung, aus der sich einige allgemeinere Folgerungen ableiten lassen.

			Ti = 50	Zr = 90	? = 180
			V = 51	Nb = 94	Ta = 182
			Cr = 52	Mo = 96	W = 186
			Mn = 55	Rh = 104,4	Pt = 197,4
			Fe = 56	Ru = 104,4	Ir = 198
		Ni = 59	Co = 59	Pd = 106,6	Os = 199
H = 1			Cu = 63,4	Ag = 108	Hg = 200
	Be = 9,4	Mg = 24	Zn = 65,2	Cd = 112	
	B = 11	Al = 27,4	? = 68	Ur = 116	Au = 197?
	C = 12	Si = 28	? = 70	Sn = 118	
	N = 14	P = 31	As = 75	Sb = 122	Bi = 210?
	O = 16	S = 32	Se = 79,4	Te = 128?	
	F = 19	Cl = 35,5	Br = 80	J = 127	
Li = 7	Na = 23	K = 39	Rb = 85,4	Cs = 133	Tl = 204
		Ca = 40	Sr = 87,6	Ba = 137	Pb = 207
		? = 45	Ce = 92		
		?Er = 56	La = 94		
		?Yt = 60	Di = 95		
		?In = 75,6	Th = 118?		

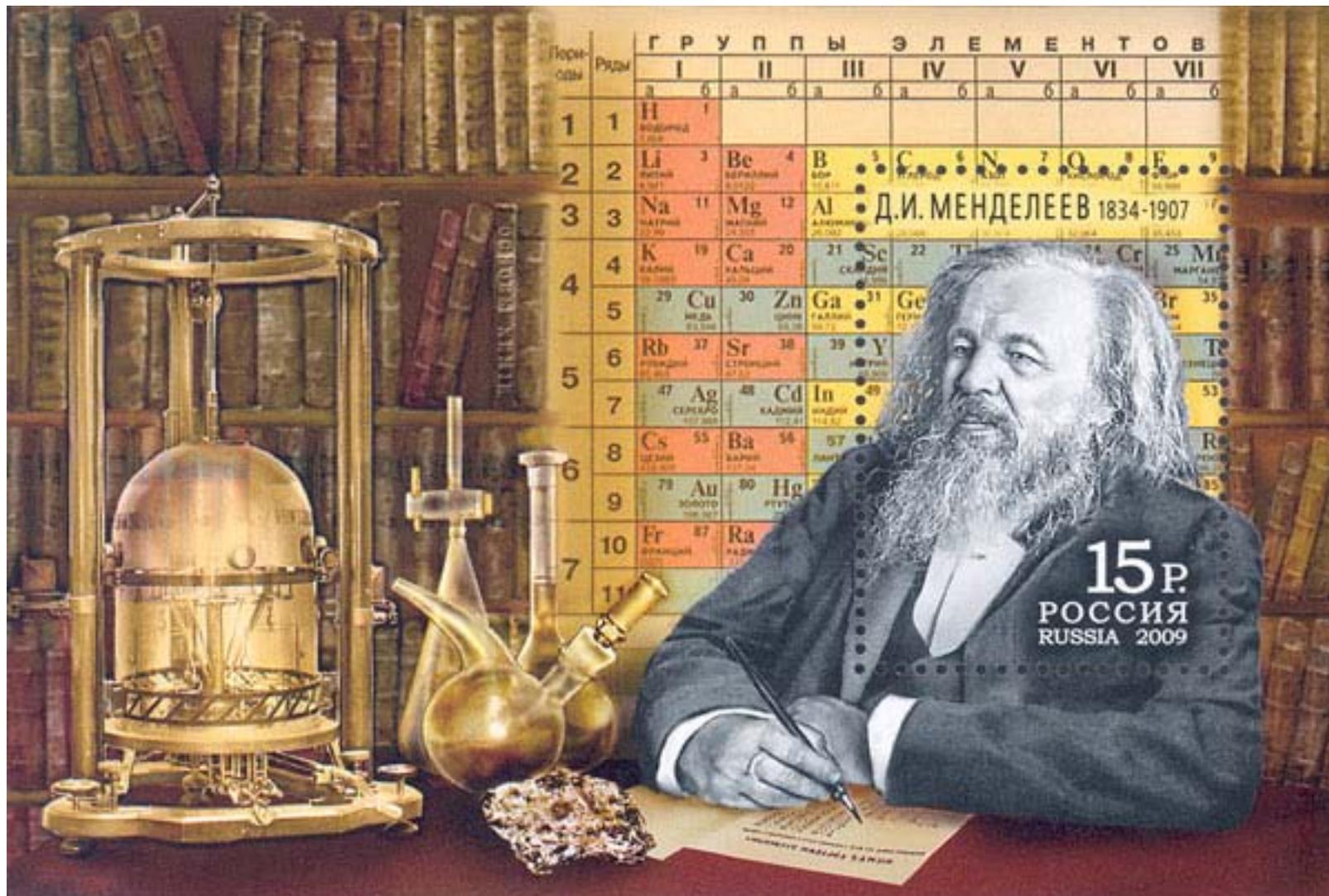
The elements Sc, Ga and Ge were all discovered within a few years...

100th Anniversary of Mendeleev's Death (2007)

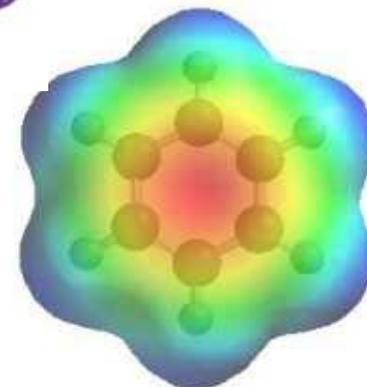
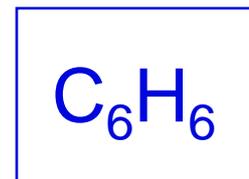
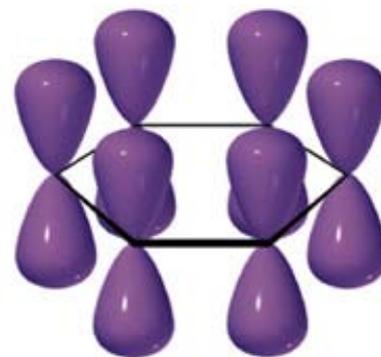
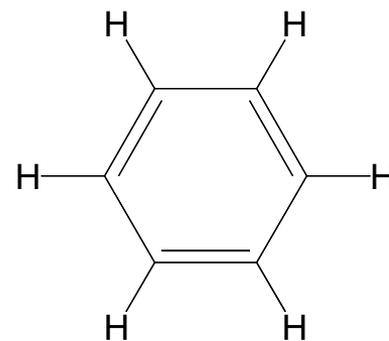


Design: Javier García-Martínez

175th Anniversary of Mendeleev's Birth (February 6, 2009)

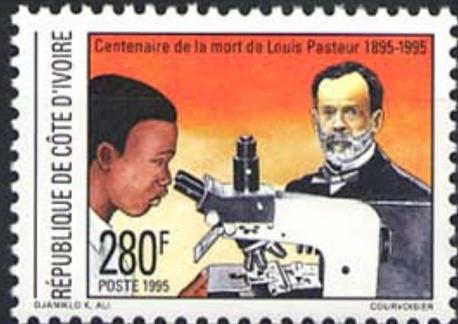


Friedrich August Kekulé (1829-1896)

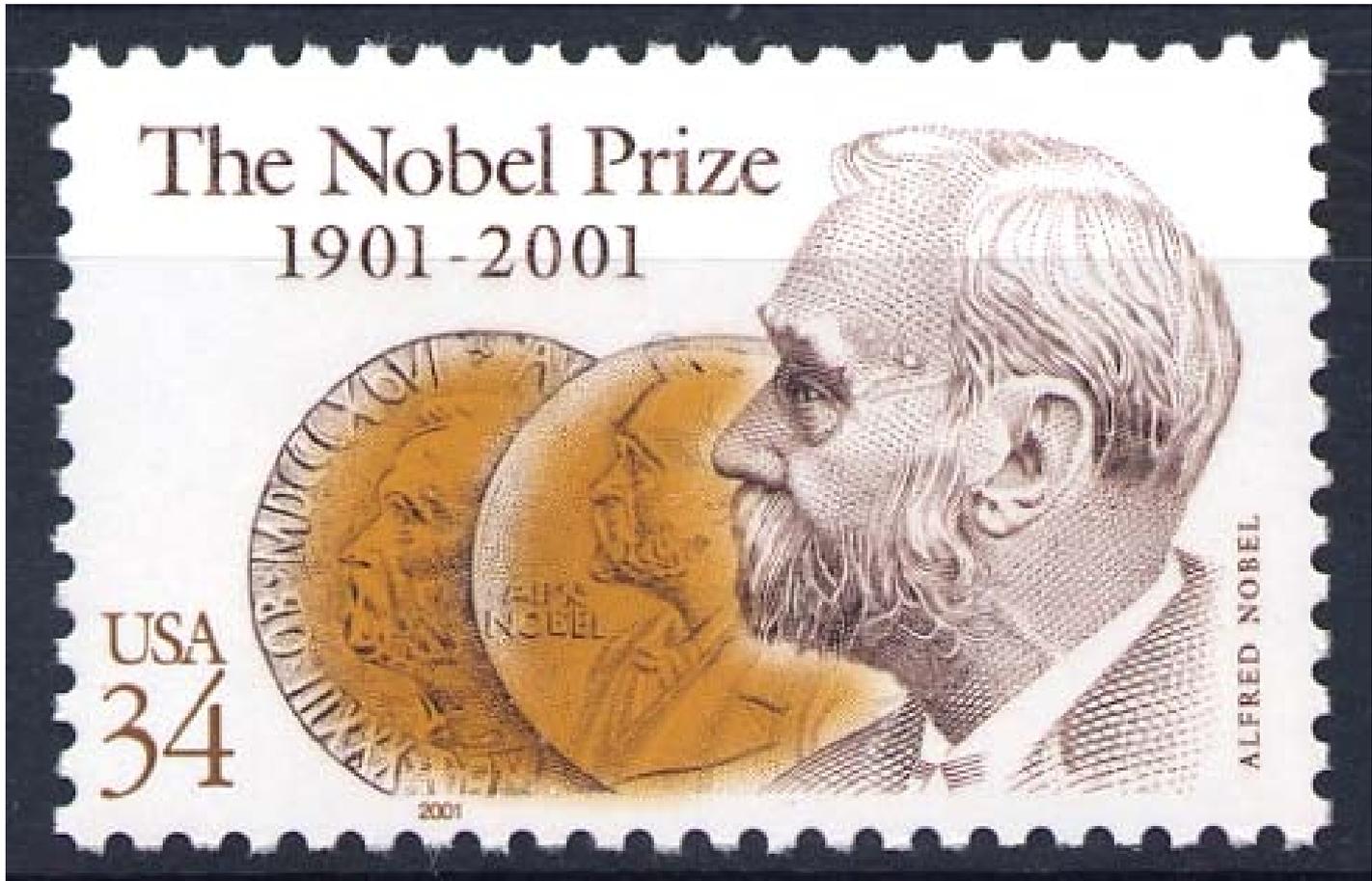


Louis Pasteur (1822-1895)





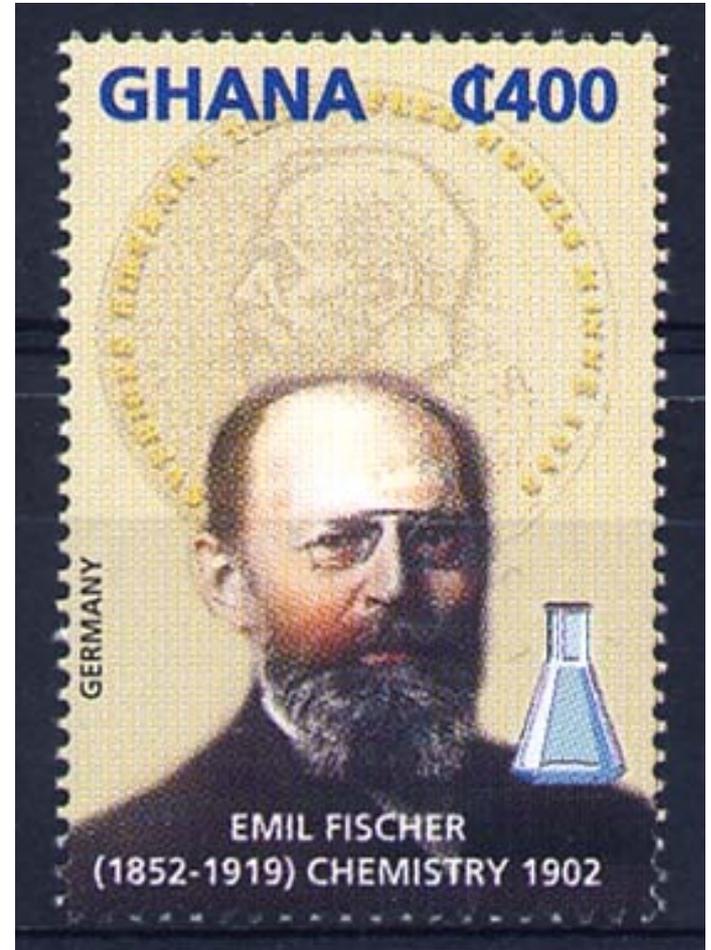
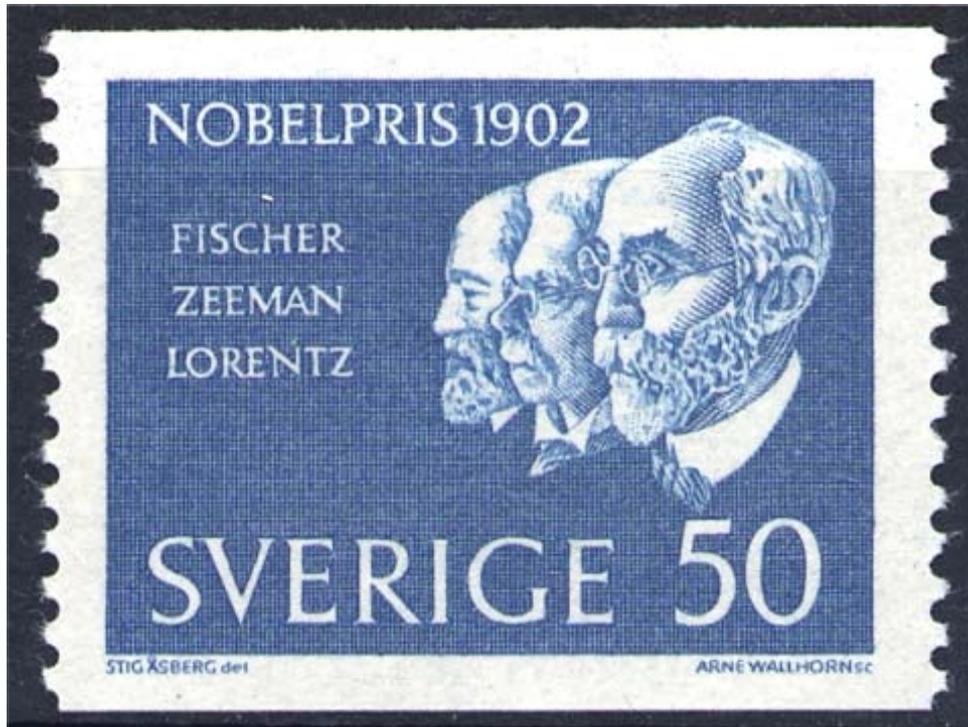
Alfred Nobel (1833-1896)

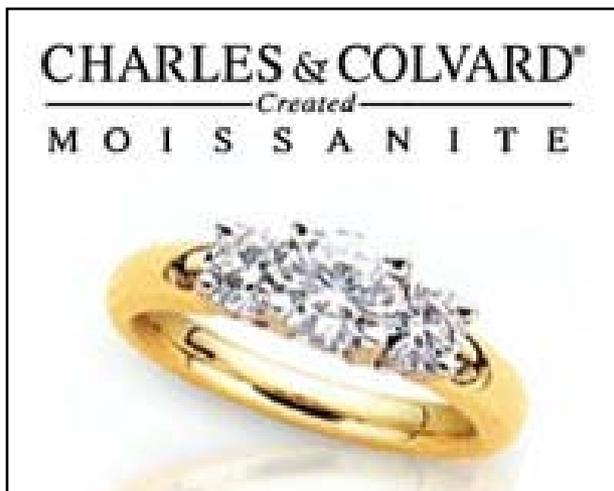




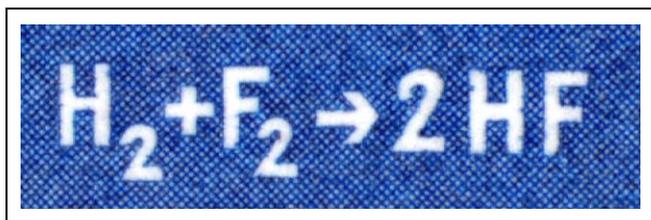
Emil Fischer (1852-1919)

“...for his work on sugar and purine syntheses”





Henri Moissan (1852-1907)



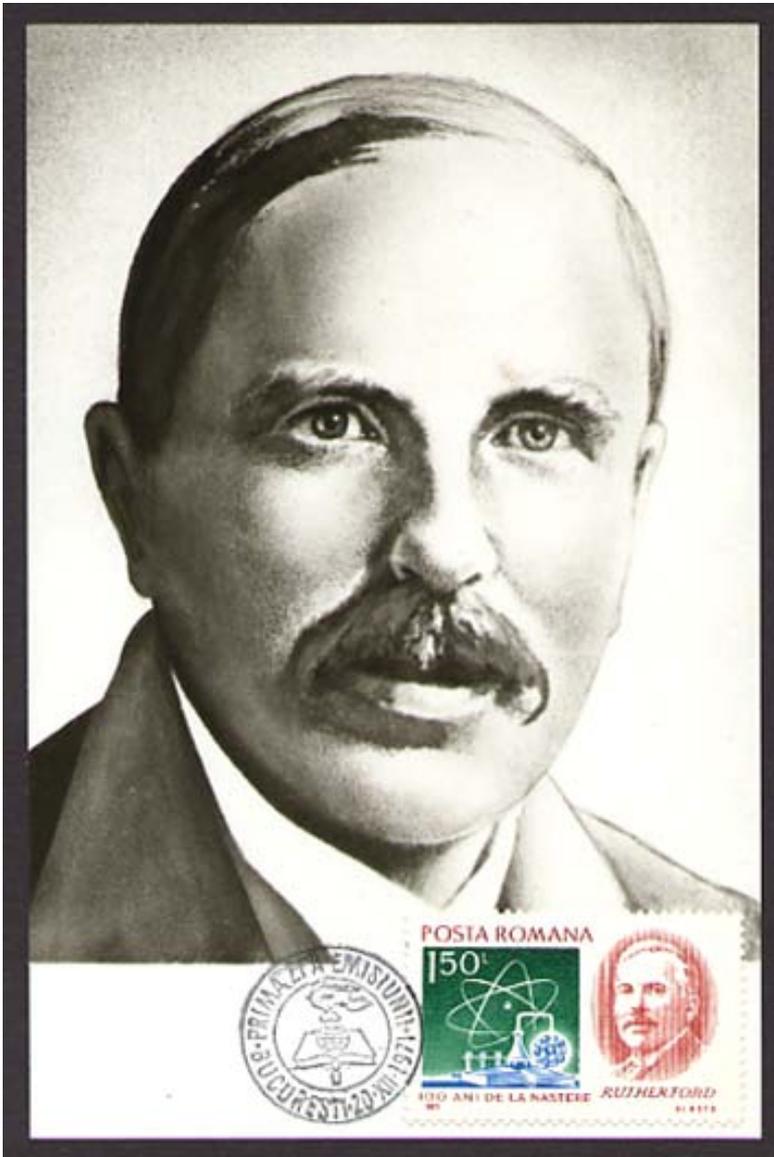
“...for his... isolation of the element fluorine... and his work on the electric furnace named after him”

Ernest Rutherford
(1871-1937)



(Chemistry, 1908)

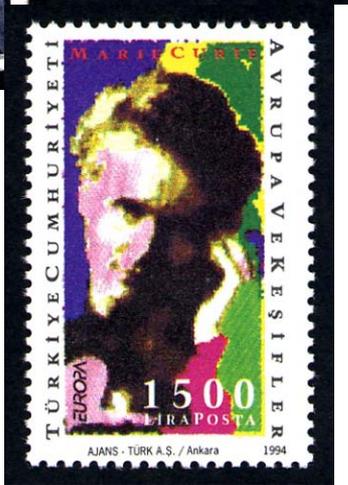
“The only science is physics.
All the rest is stamp collecting.”



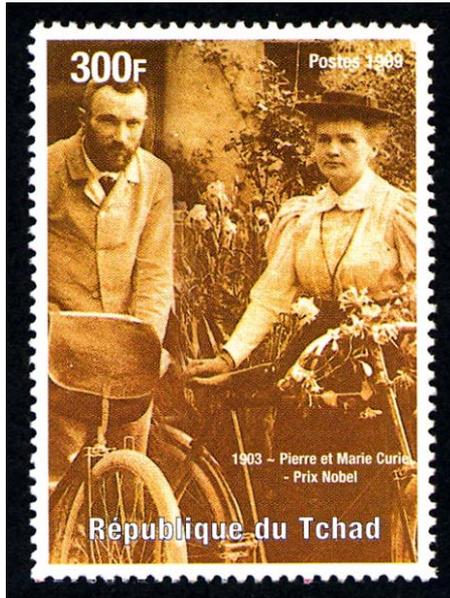
Marie Curie (1867-1934)



[Physics, 1903 & Chemistry, 1911]



Marie and Pierre Curie



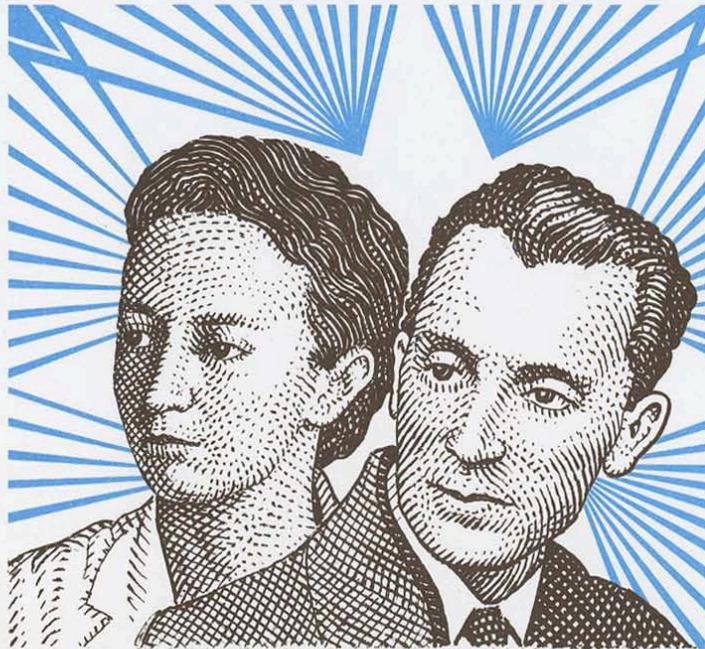
Irène and Frédéric Joliot-Curie



Nobel Prize in Chemistry (1935)

“for their synthesis of new radioactive elements”

IRÈNE ET FRÉDÉRIC JOLIOT-CURIE



S. Gandon
Joliot
Langevin

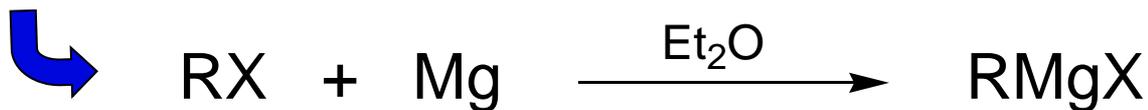
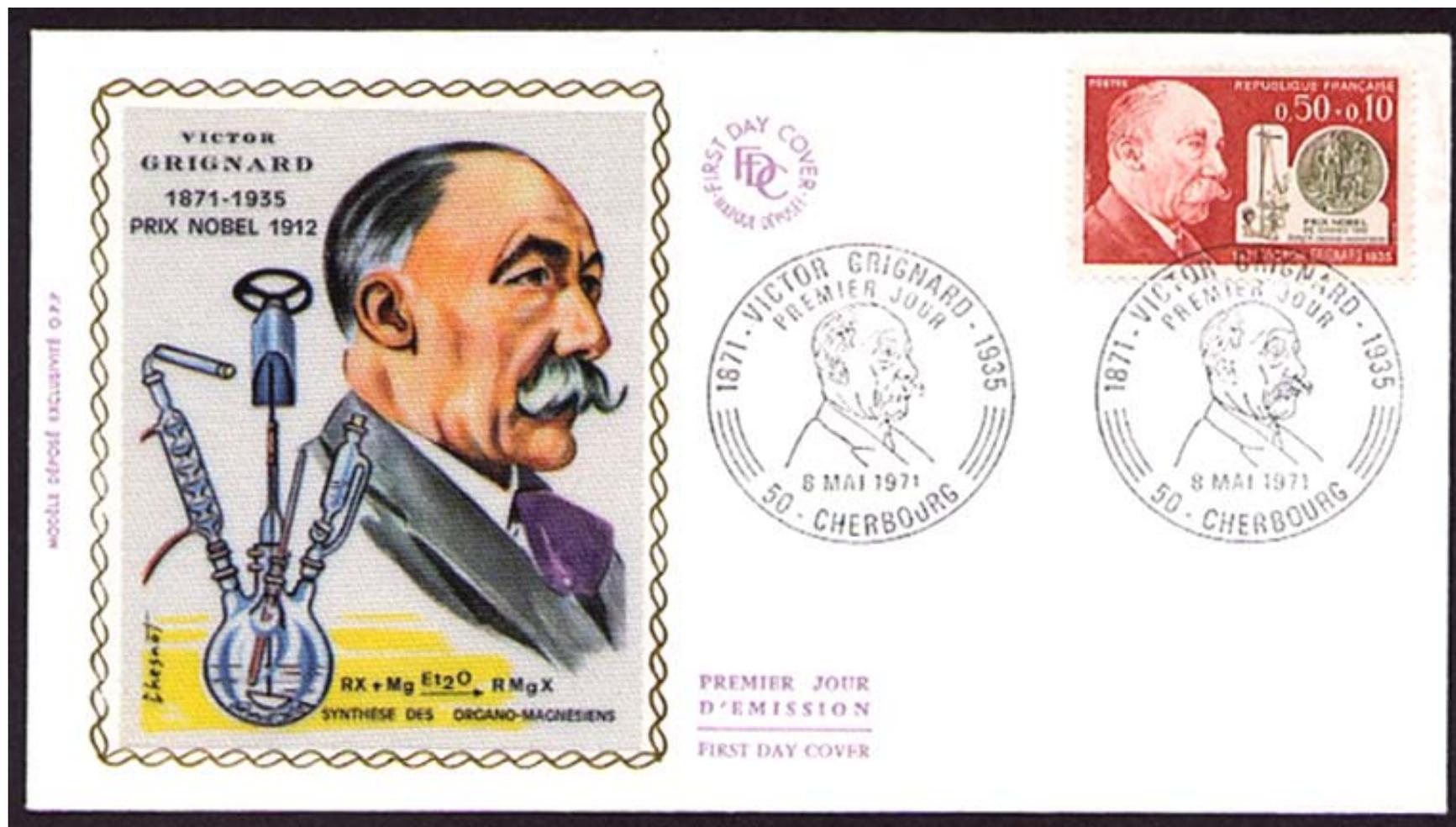


Pierre Joliot
(1932–)

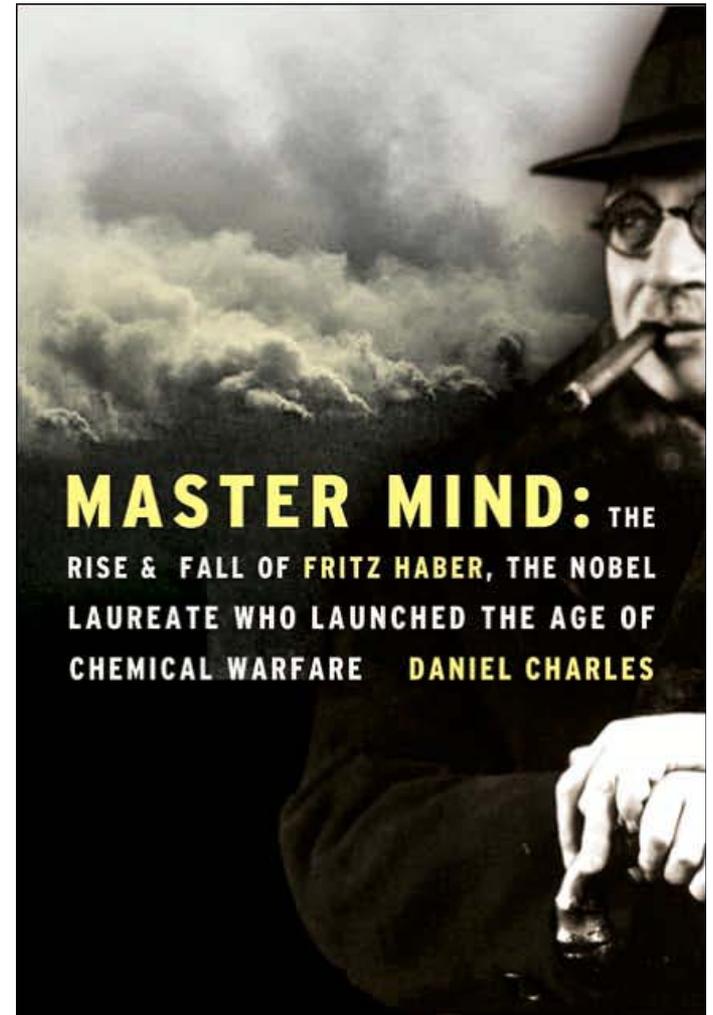


Hélène Langevin-Joliot
(1927–)

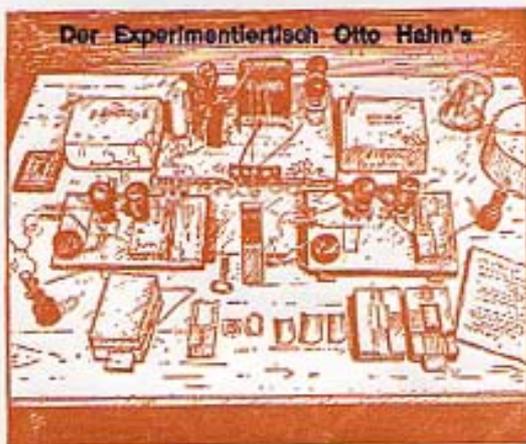
Victor Grignard (1871-1935)



Fritz Haber (1868-1934)

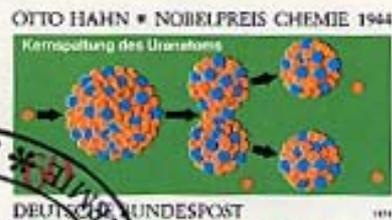


Otto Hahn (1868-1934)



Sonderpostwertzeichen
Nobelpreisträger
Otto Hahn
Ersttagsbrief

Nobelpreis Chemie 1944
Kernspaltung des Uranatoms



13048 *

Chemistry (1944): discovery of nuclear fission

Lise Meitner (1878-1968)



Karl Ziegler (1898-1973) & Giulio Natta (1903-1979)



Dorothy Crowfoot Hodgkin (1910-1994)

Chemistry Nobel '64



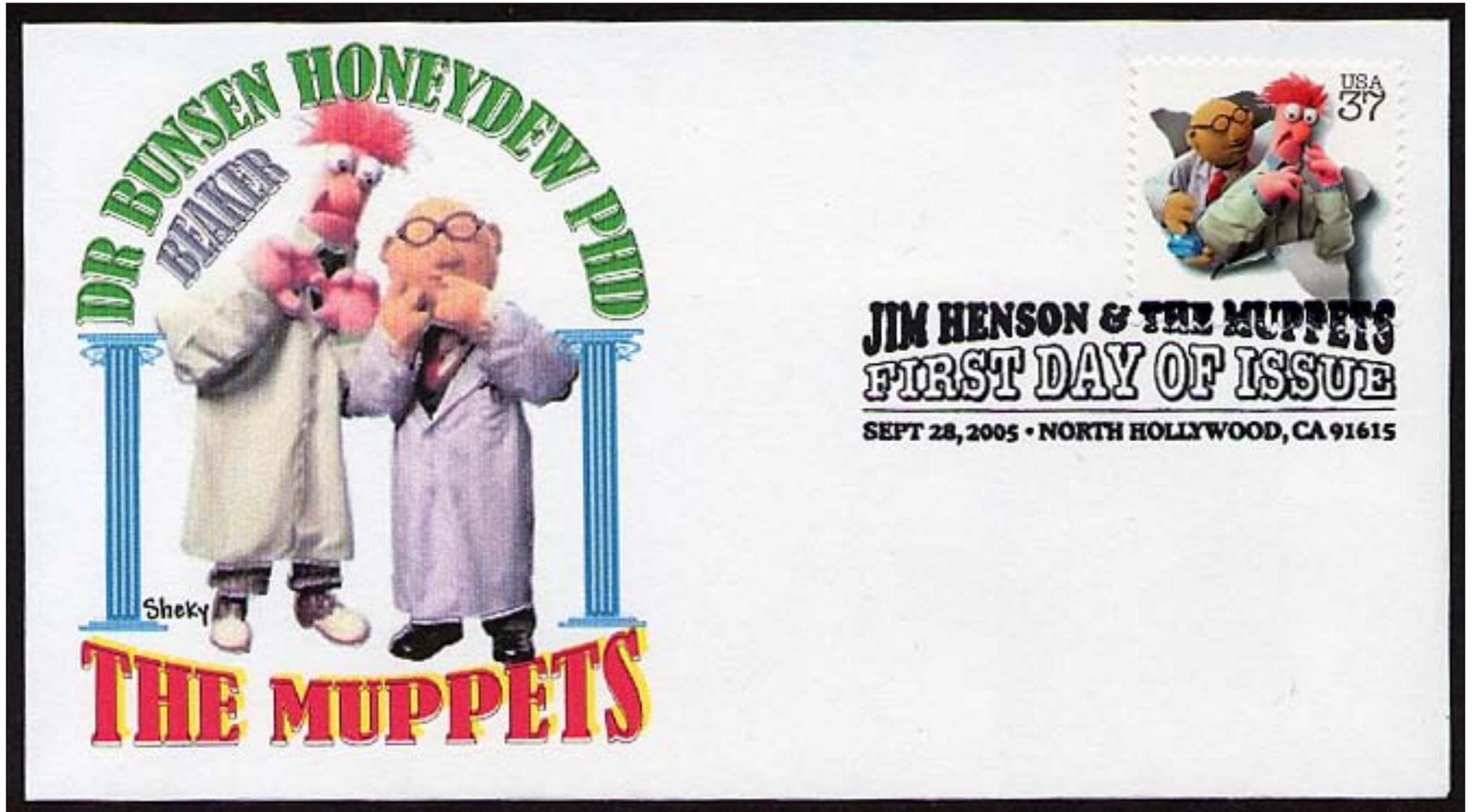
Issued on 25 February 2010: 350th anniversary of the Royal Society.

Mario J. Molina (1943–)



1995 Nobel Prize in Chemistry “for his work in atmospheric chemistry, particularly the formation and decomposition of ozone”

Dr. Bunsen Honeydew and Beaker



Atoms, molecules, minerals...

Atoms & Nuclear Energy



BUREAU INTERNATIONAL DES POIDS ET MESURES FRANCE

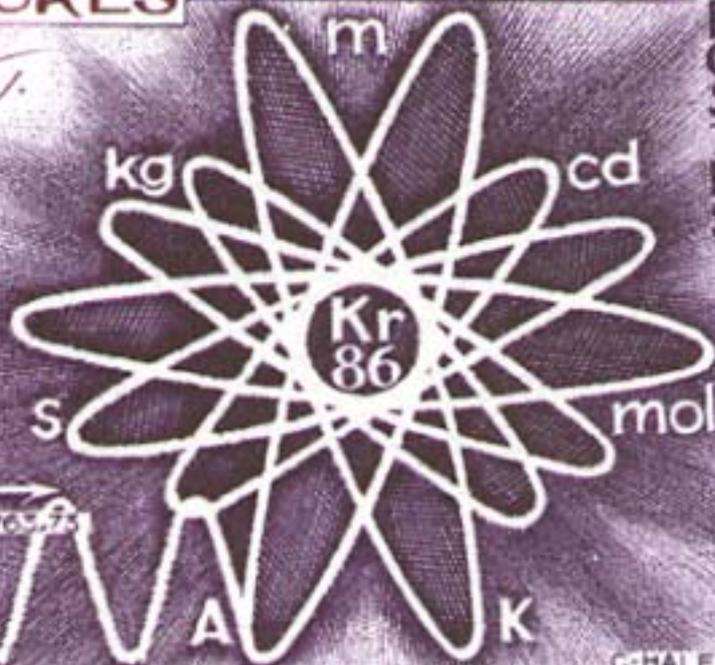
Fait à Paris le 20 Mai 1875.



*Hecard
C. de la Roche
Wamag*

Wohlenbutz

100

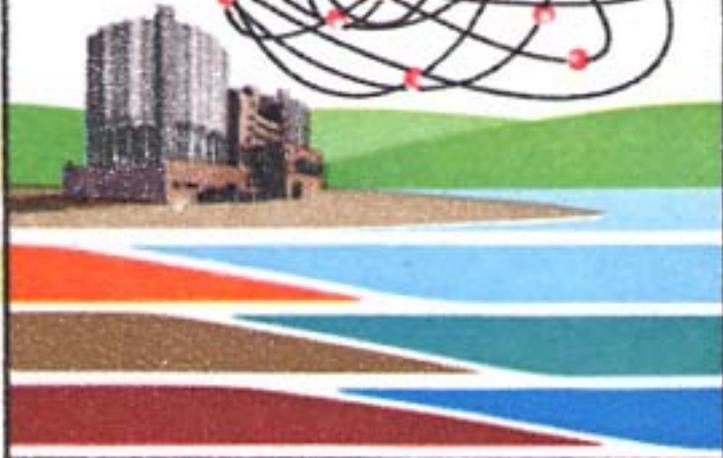
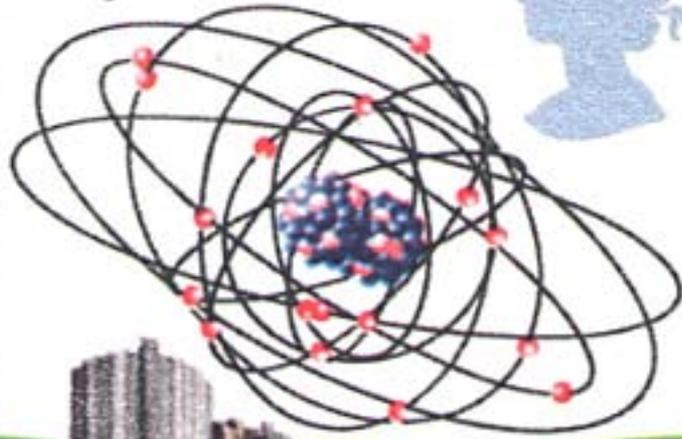


POSTALIS

CHALEY

1875 ~ CONVENTION DU METRE ~ 1975

13p



ELECTRICITY

Elements & Minerals





2nd



CHEMISTRY

Nobel Prize 100th Anniversary

Tristan Da Cunha



SULPHUR

10p

SLOVENIJA



Cinabarit Idrija

80



Hg

M. UČAKAR - M. UDOVČ

1999

DELO - TISKARNA



1975
CORREOS



NIQUEL

CUBA MINERALES CUBANOS 3

C A N A D A

GOLD/OR
Au



4 2

ZAÏRE

2K



MALACHITE

1983

COURVOISIER

ZAÏRE

6Z



DIOPTASE

1983

COURVOISIER

ZAÏRE

8Z



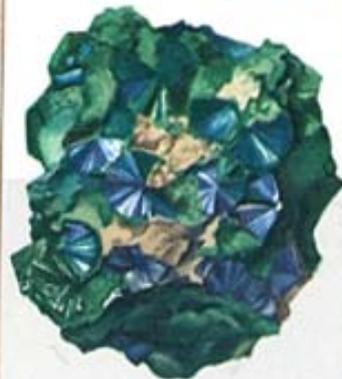
CUPRITE

1983

COURVOISIER

ZAÏRE

1.50Z



BOURNONITE

1983

COURVOISIER

ZAÏRE

45K



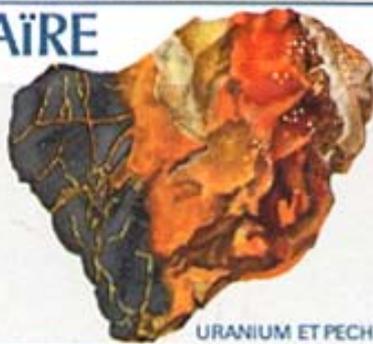
QUARTZ

1983

COURVOISIER

ZAÏRE

1Z



URANIUM ET PECHBLENDE

1983

COURVOISIER

ZAÏRE

3Z



CASSITERITE

1983

COURVOISIER

stibnite

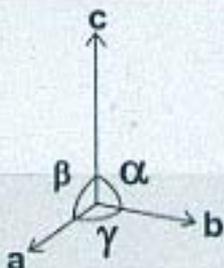
cinnabar

wolframite



PERÚ

SI. 180



$a = b \neq c$
 $\alpha = \beta = \gamma = 90^\circ$

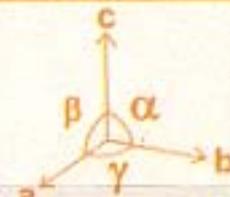
CALCOPIRITA

2001

CuFeS2

PERÚ

SI. 320



$a = b = c$
 $\alpha = \beta = \gamma = 90^\circ$

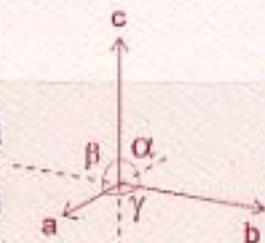
ESFALERITA

2001

ZnS

PERU

SI. 270



$a = b = c$
 $\alpha = \beta = \gamma = 90^\circ$

GALENA

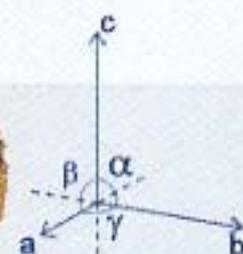
1999

THOMAS GREGG & SONS

PbS

PERU

SI. 330



$a = b \neq c$
 $\alpha = \beta = \gamma = 90^\circ$

SCHEELITA

1999

THOMAS GREGG & SONS

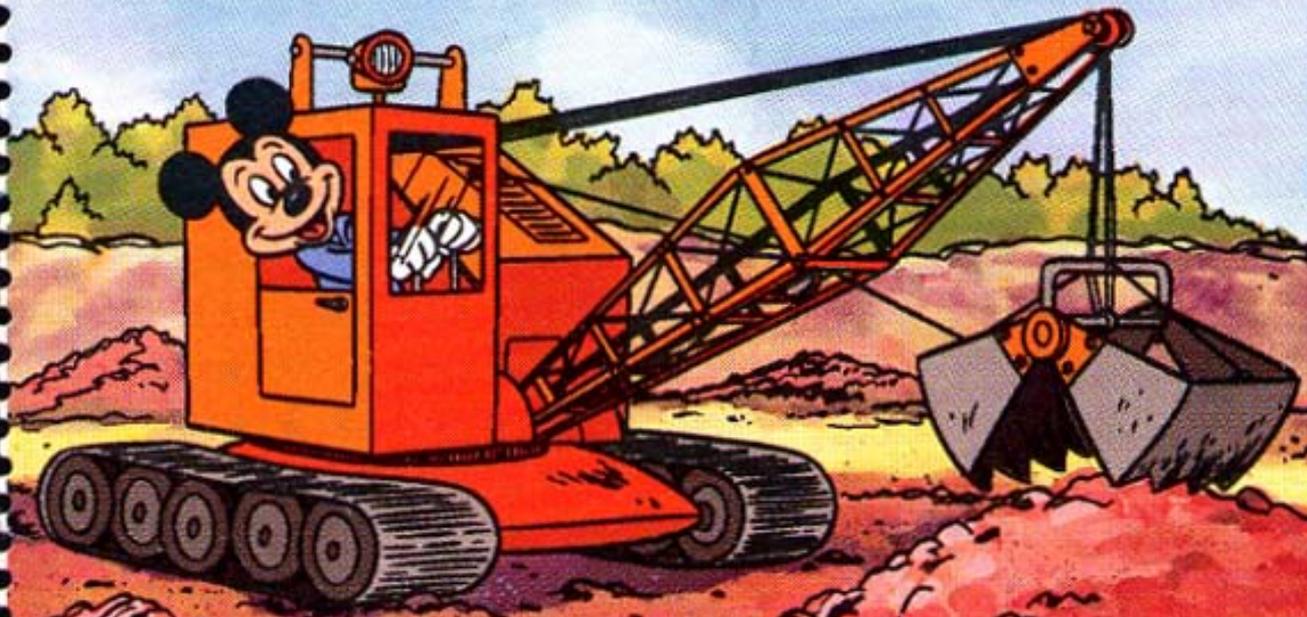
CaWO4

Aluminum and Bauxite on Stamps...



SIERRA LEONE

Le 16



© DISNEY

MICKEY MINING BAUXITE

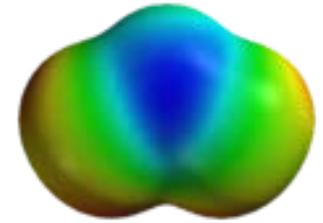
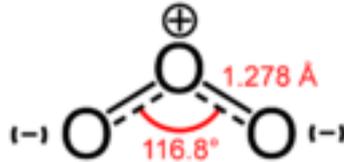
Chemical Symbols & Formulas



*International Union of Pure and Applied Chemistry
20th Congress: Moscow, 1965*

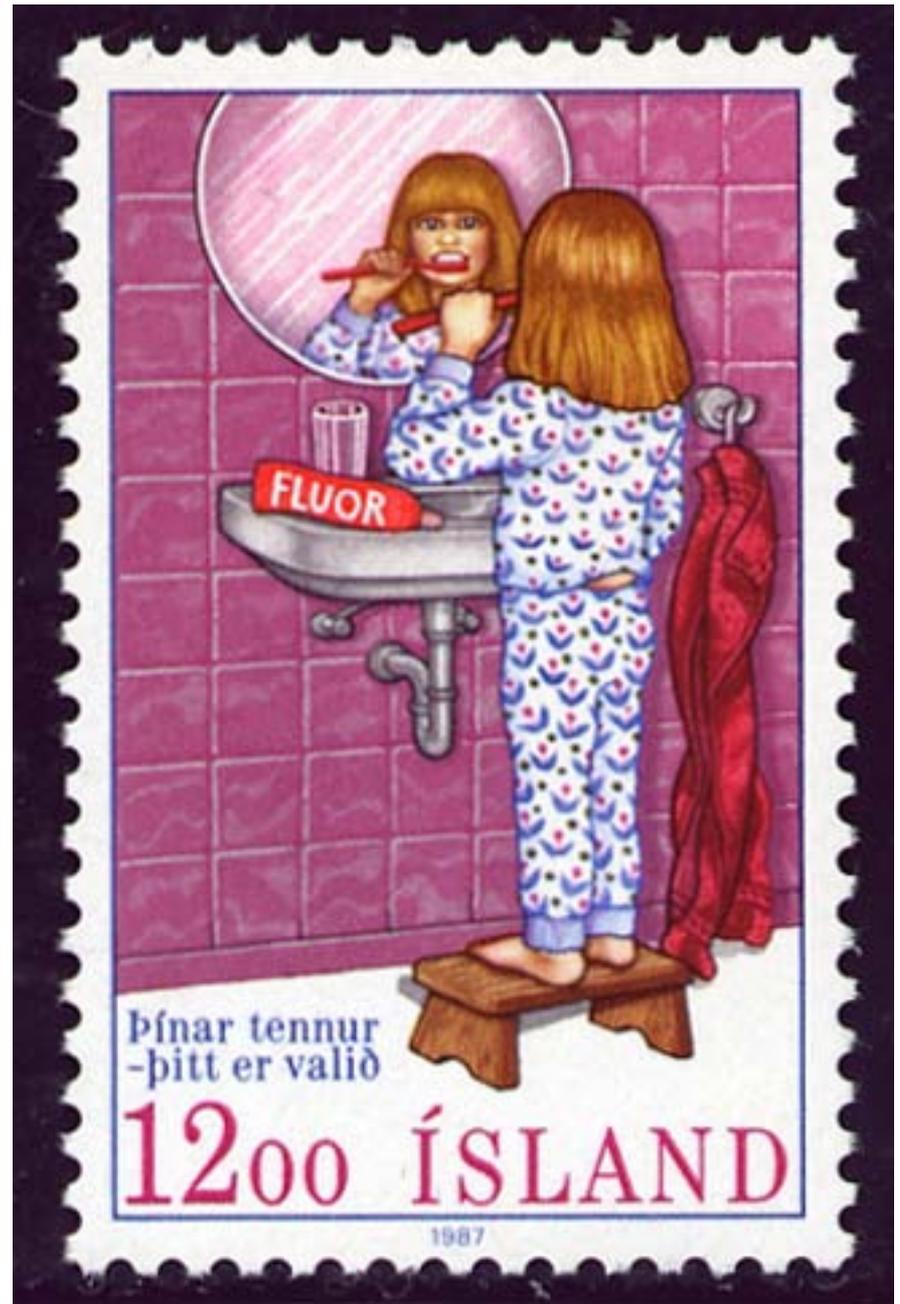


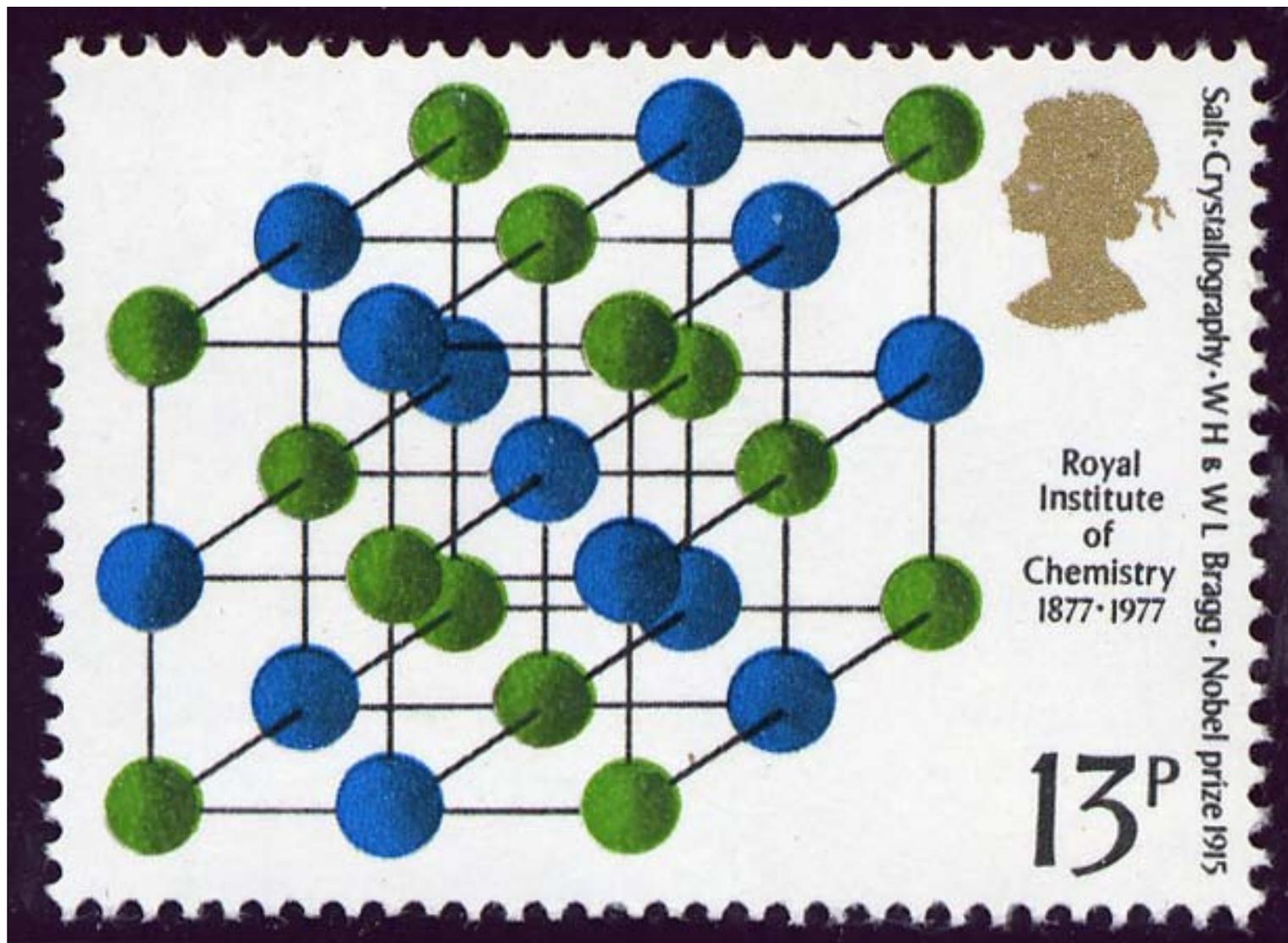
Ozone, O_3



C. F. Schönbein (1799-1868)

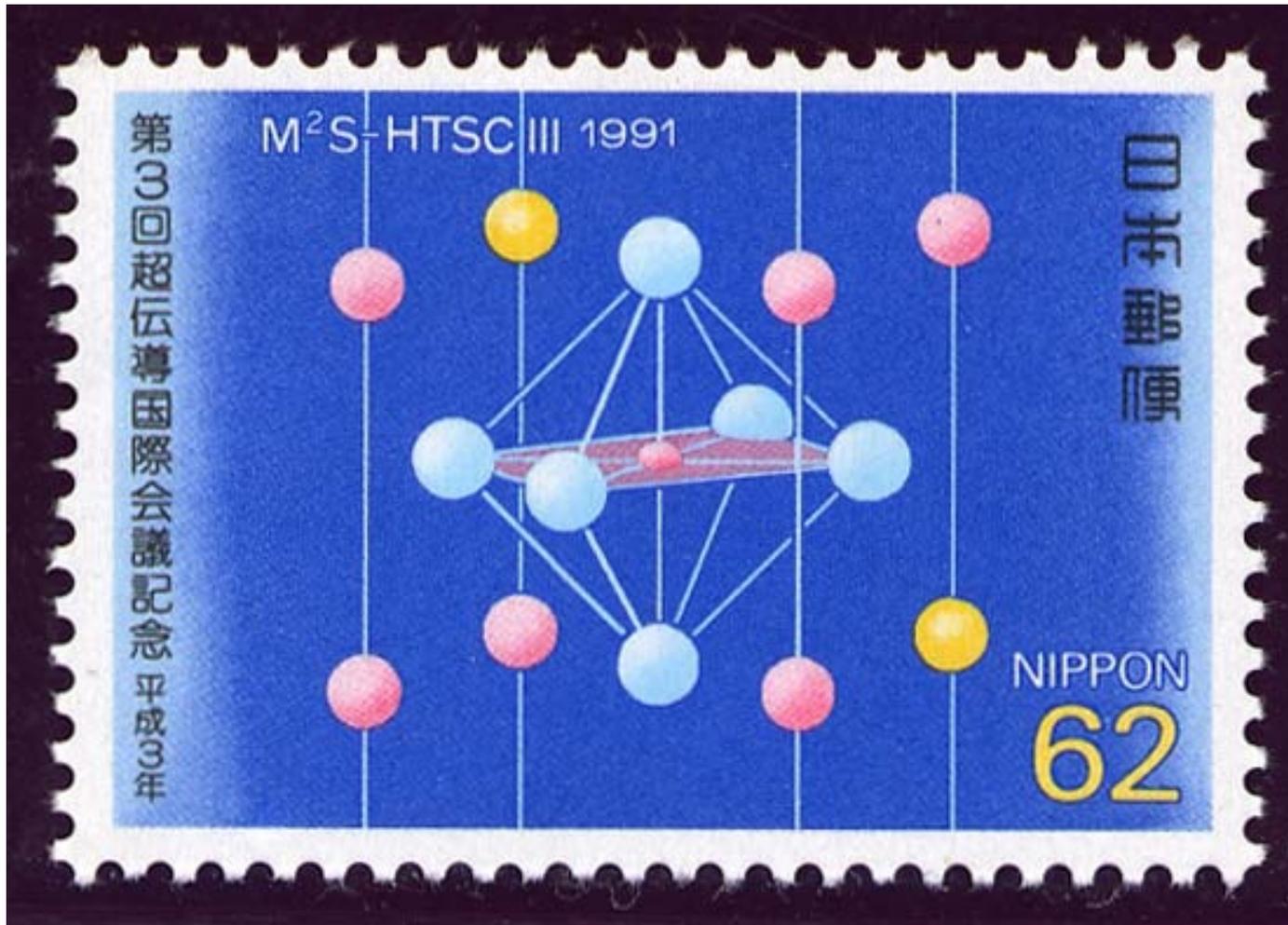
Fluoride & toothpaste



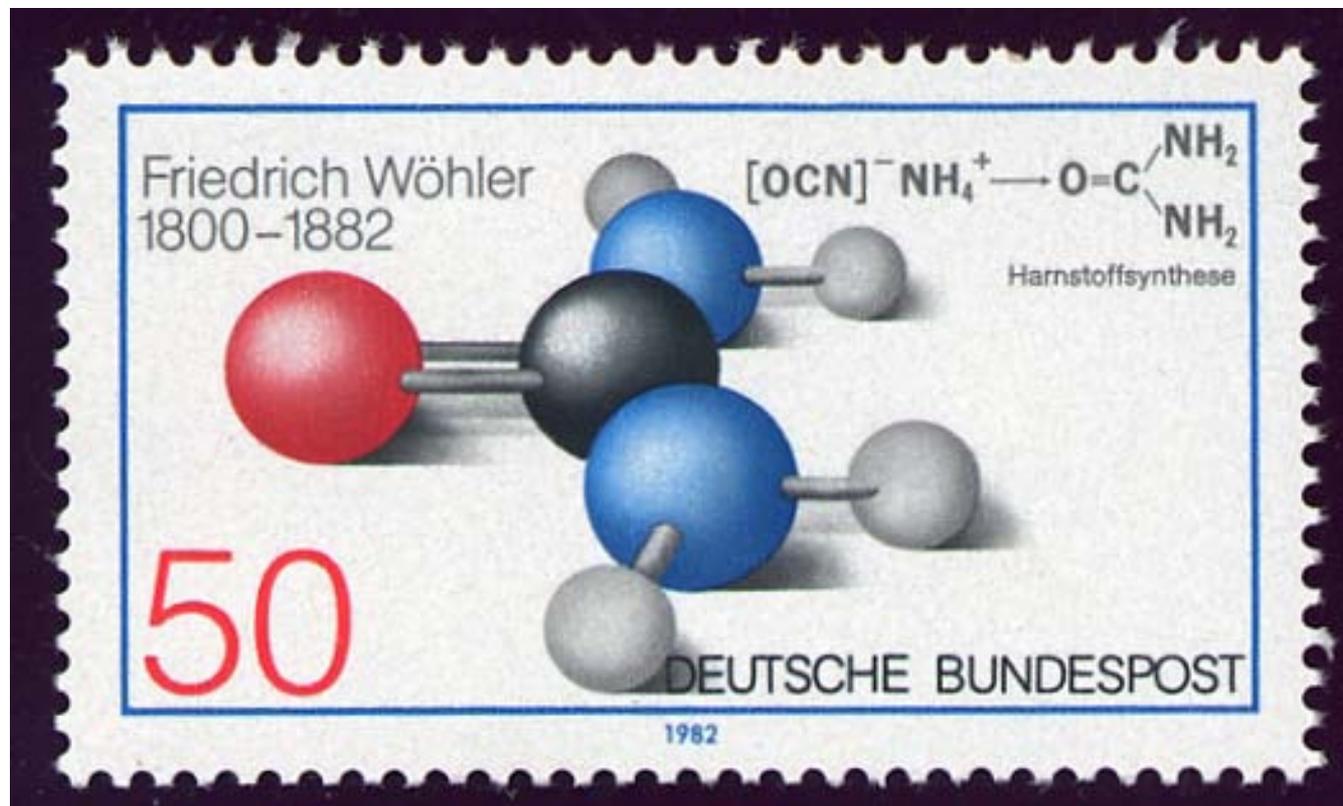


NaCl

High Temperature SuperConductors



Friedrich Wöhler and the synthesis of urea

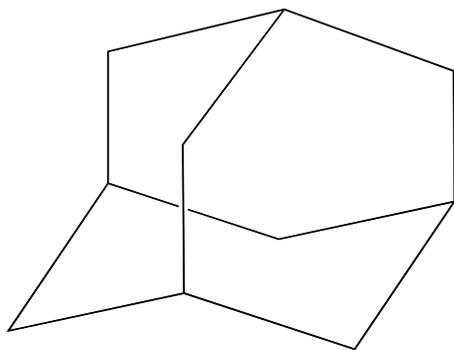


- First isolation of pure aluminum.
- Co-discoverer of beryllium, silicon, titanium and yttrium.
- Synthesis of acetylene from calcium carbide.



Vladimir Prelog (1906-1998)
Chemistry Nobel '75

adamantane



See: *Chemistry International* **2008**, 30 (1), 7.

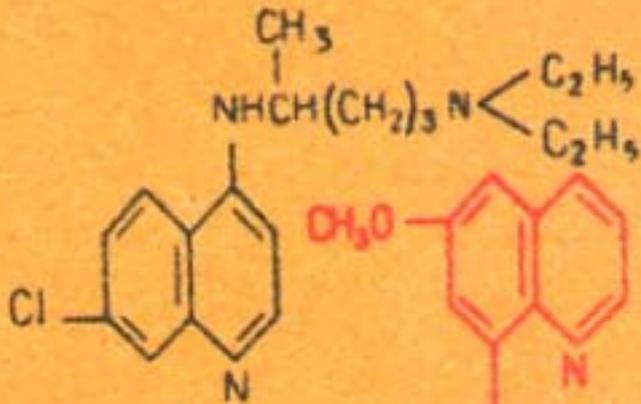
2,2-difluorobutane



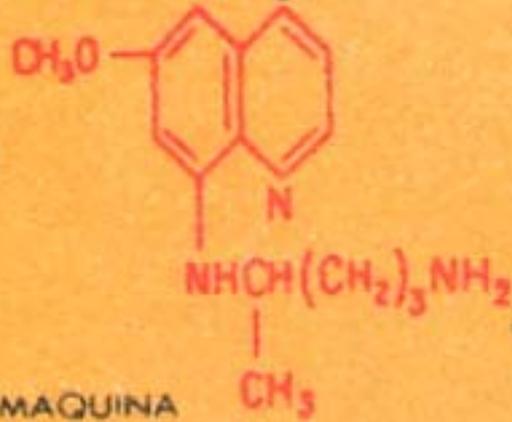
100th anniversary of Geneva Conference (1892)

EL MUNDO UNIDO CONTRA LA MALARIA

CLOROQUINA



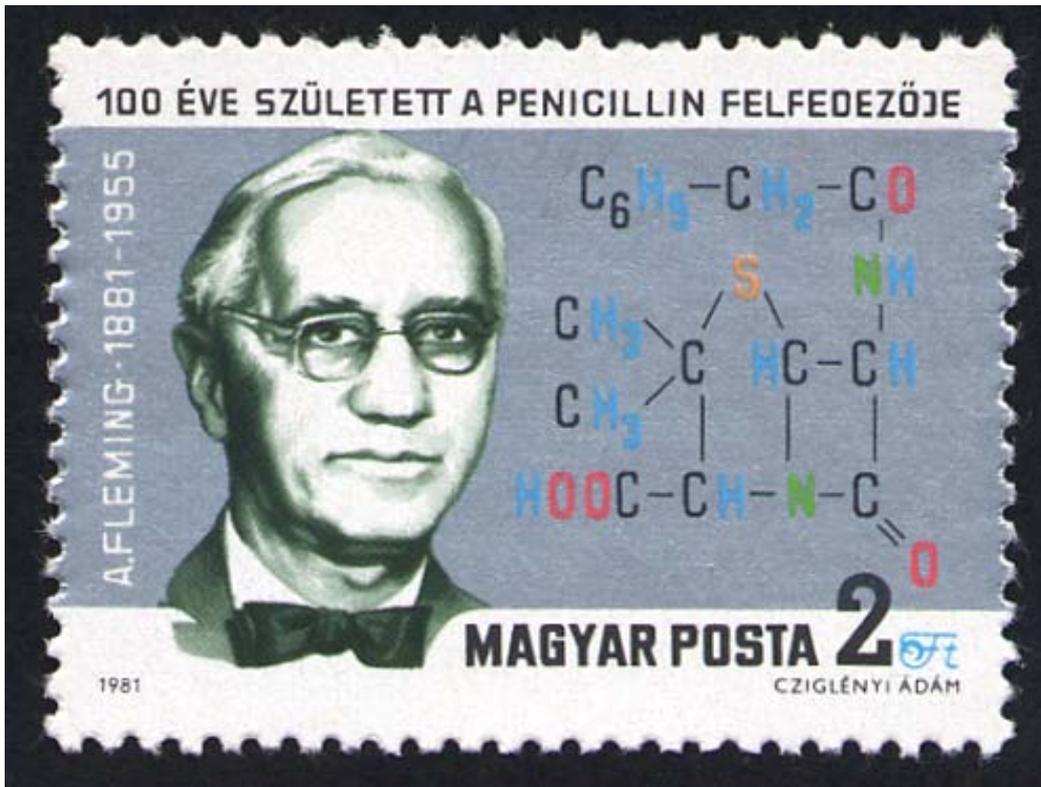
Chinchona



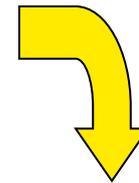
PRIMAQUINA

3 ¢

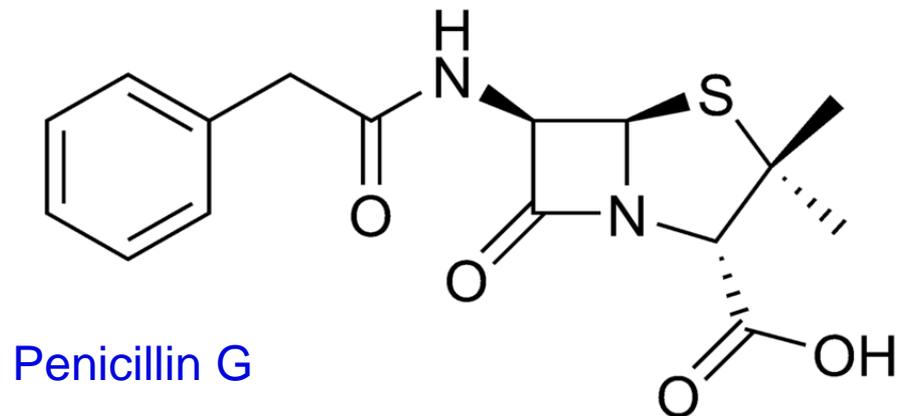
CORREOS DE CUBA



Alexander Fleming
(1881-1955)



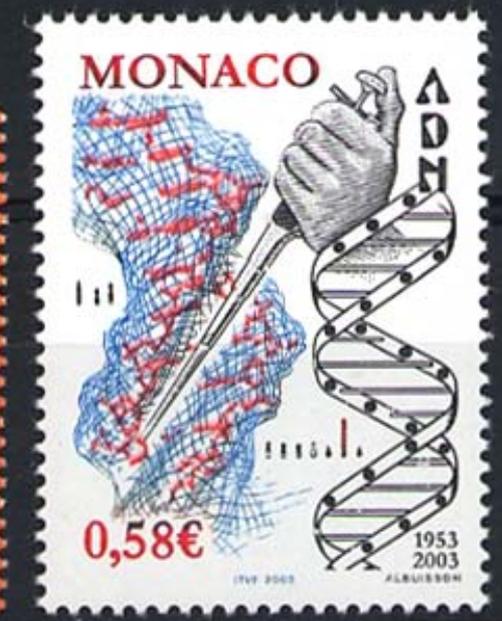
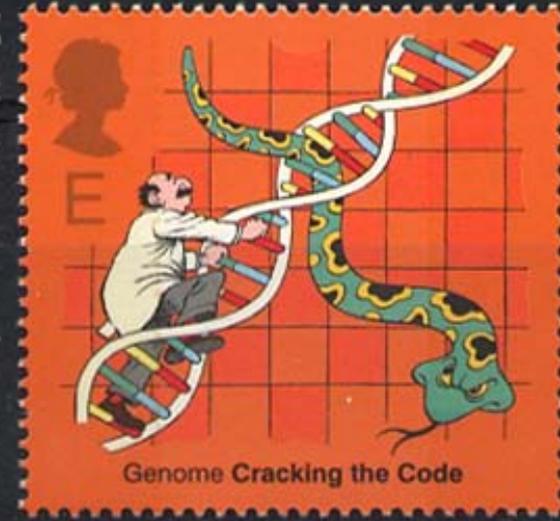
(Medicine, 1945)



Biochemistry...

The DNA double helix





Base pairs in DNA: $G \Leftrightarrow C$ and $A \Leftrightarrow T$



Glassware...

Glassware





Chemistry students and chemists...

Chemistry Students



International
Year
of the Child
1979

MAURITIUS R1.50

1966 - INAUGURATION NOUVEAU SIEGE O. M. S.



RÉPUBLIQUE DE GUINÉE 30 F

Chemists



Chemical industry...

Chemical Industry



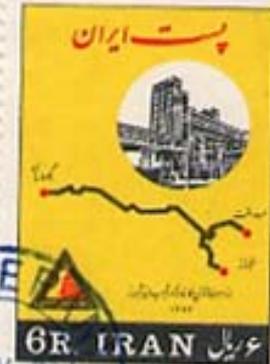
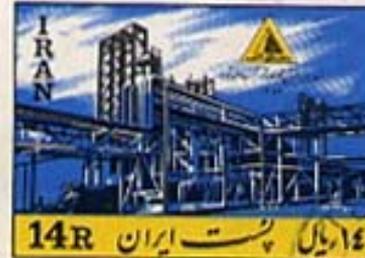
Fertilizers

FIRST DAY COVER



بمناسبت افتتاح کارخانه کود شیمیایی شیراز
۲۶ مهر ماه ۱۳۴۳

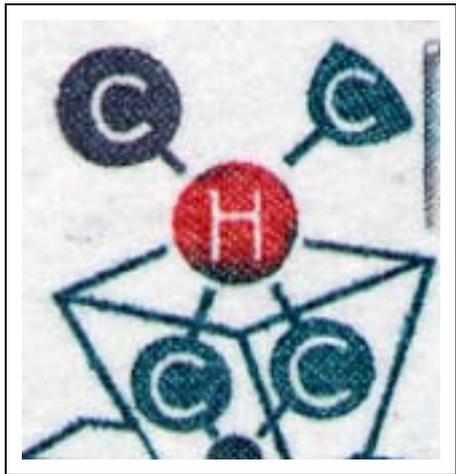
COMMEMORATING
INAUGURATION OF THE CHEMICAL FERTILIZER
PLANT IN SHIRAZ
18th October, 1965



Petrochemistry



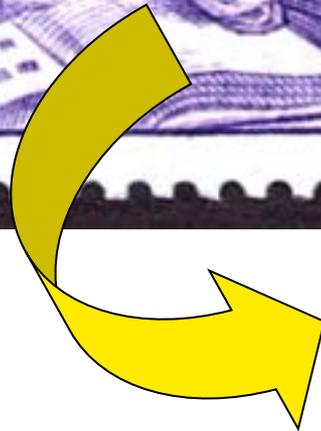
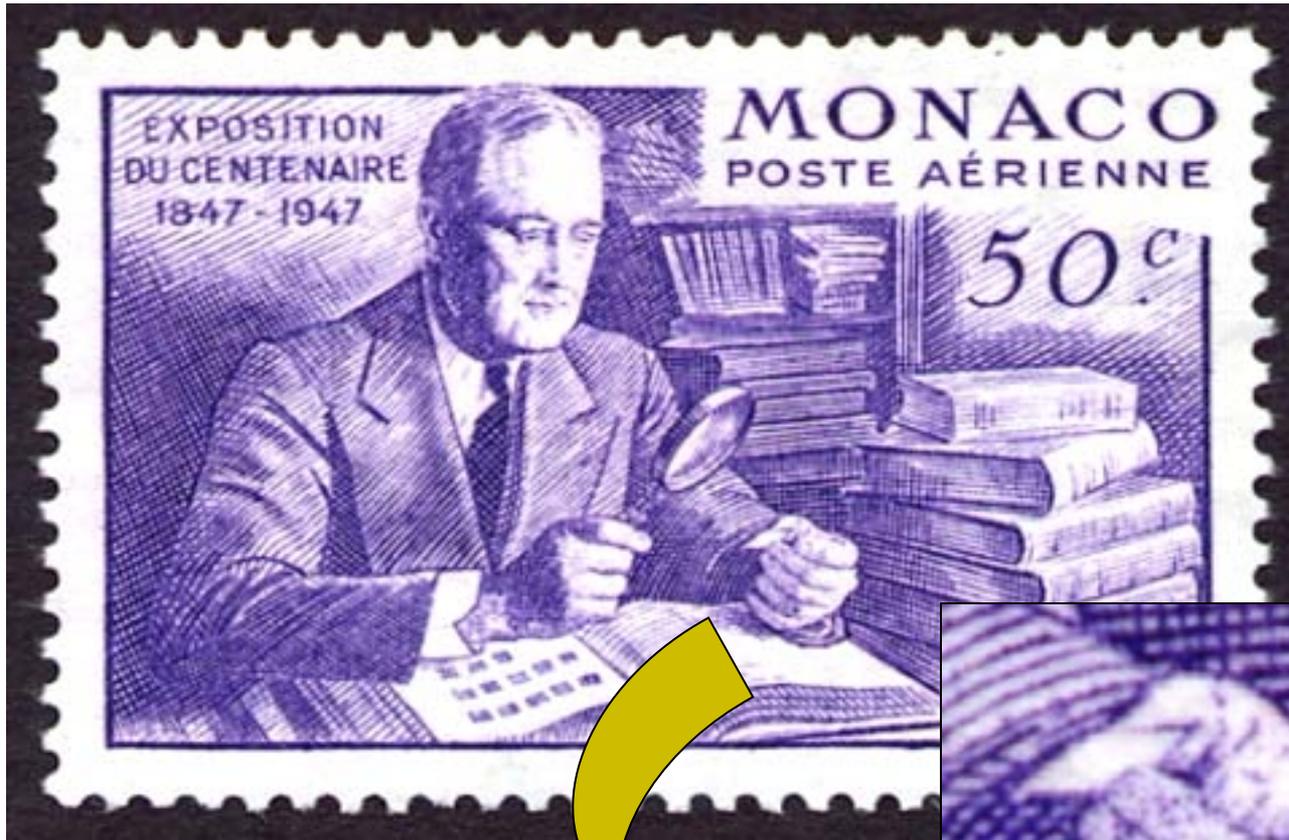
Plastics



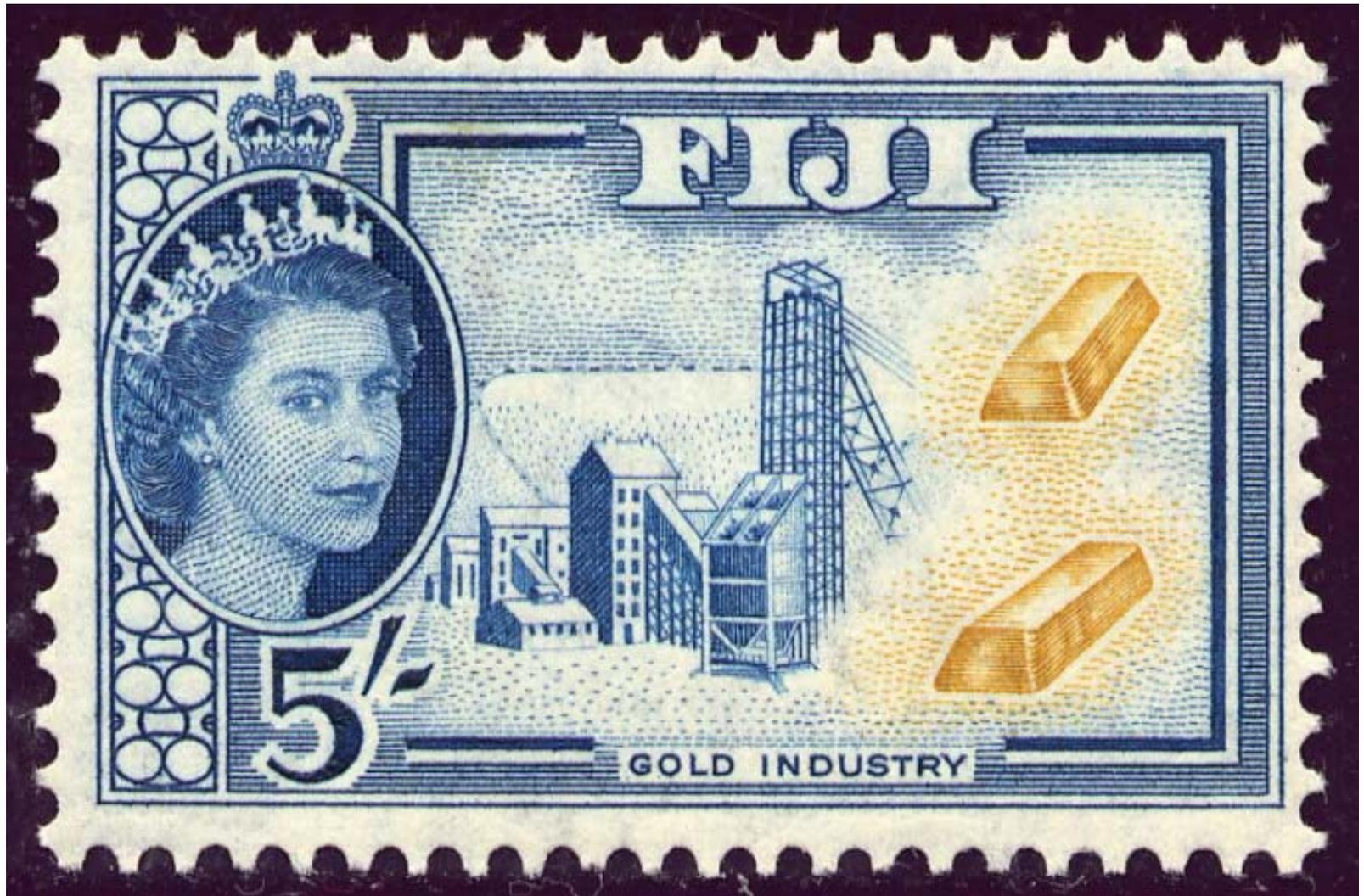
methane: CH_4



FDR, stamp collector



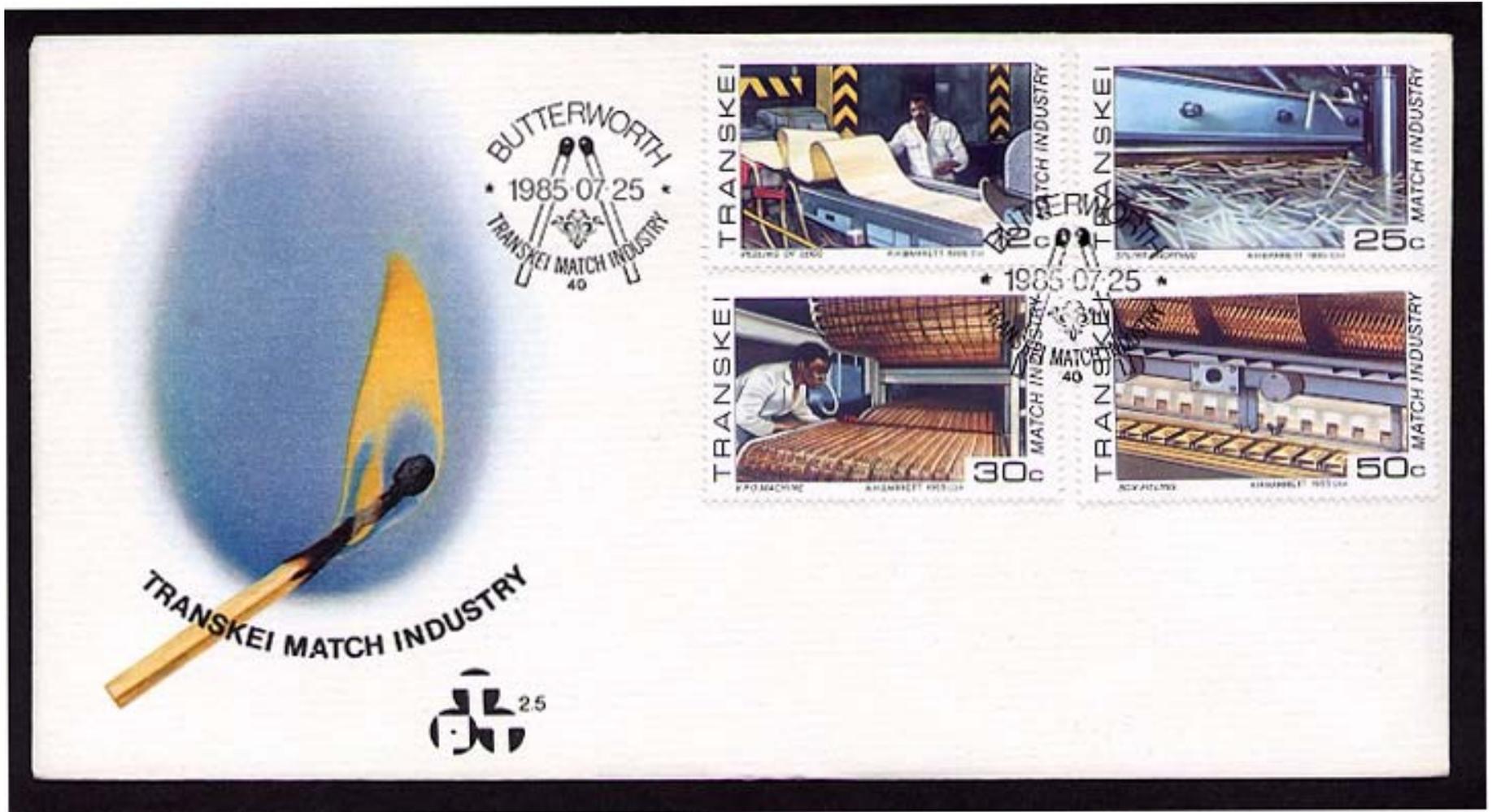
Gold



Salt



Matches



And a few special items...

Hydrogen balloons...

May 6th, 1936

Via Airship "HINDENBURG"

from: FRANKFURT / MAIN, GERMANY

to: LAKEHURST, N. J., U. S. A.



VIA AIR MAIL

EINSCHREIBEN

Gus Nichols
R. 3 Box 254
Petaluma
California

U. S. A.



*The crash of the "Hindenburg" in Lakehurst, NJ
May 6th, 1937*



Element 112: Copernicium

- Name proposed by Sigurd Hofmann *et al.* (Center for Heavy Ion Research) was accepted at the 45th General Assembly of IUPAC (Aug. '09)
- The proposed chemical symbol (Cp) was initially rejected... and then changed to Cn.



2011 Periodic Table of the Elements (IUPAC)

IUPAC Periodic Table of the Elements

Key:

atomic number
Symbol
name
standard atomic weight

1 H hydrogen [1.007, 1.009]																	18 He helium 4.003						
3 Li lithium [6.938, 6.997]	4 Be beryllium 9.012																	5 B boron [10.80, 10.83]	6 C carbon [12.00, 12.02]	7 N nitrogen [14.00, 14.01]	8 O oxygen [15.99, 16.00]	9 F fluorine 19.00	10 Ne neon 20.18
11 Na sodium 22.99	12 Mg magnesium 24.31																	13 Al aluminium 26.98	14 Si silicon [28.08, 28.09]	15 P phosphorus 30.97	16 S sulfur [32.05, 32.08]	17 Cl chlorine [35.44, 35.48]	18 Ar argon 39.95
19 K potassium 39.10	20 Ca calcium 40.08	21 Sc scandium 44.96	22 Ti titanium 47.87	23 V vanadium 50.94	24 Cr chromium 52.00	25 Mn manganese 54.94	26 Fe iron 55.85	27 Co cobalt 58.93	28 Ni nickel 58.69	29 Cu copper 63.55	30 Zn zinc 65.38(2)	31 Ga gallium 69.72	32 Ge germanium 72.63	33 As arsenic 74.92	34 Se selenium 78.96(3)	35 Br bromine 79.90	36 Kr krypton 83.80						
37 Rb rubidium 85.47	38 Sr strontium 87.62	39 Y yttrium 88.91	40 Zr zirconium 91.22	41 Nb niobium 92.91	42 Mo molybdenum 95.96(2)	43 Tc technetium	44 Ru ruthenium 101.1	45 Rh rhodium 102.9	46 Pd palladium 106.4	47 Ag silver 107.9	48 Cd cadmium 112.4	49 In indium 114.8	50 Sn tin 118.7	51 Sb antimony 121.8	52 Te tellurium 127.6	53 I iodine 126.9	54 Xe xenon 131.3						
55 Cs caesium 132.9	56 Ba barium 137.3	57-71 lanthanoids	72 Hf hafnium 178.5	73 Ta tantalum 180.9	74 W tungsten 183.8	75 Re rhenium 186.2	76 Os osmium 190.2	77 Ir iridium 192.2	78 Pt platinum 195.1	79 Au gold 197.0	80 Hg mercury 200.6	81 Tl thallium [204, 204.4]	82 Pb lead 207.2	83 Bi bismuth 209.0	84 Po polonium	85 At astatine	86 Rn radon						
87 Fr francium	88 Ra radium	89-103 actinoids	104 Rf rutherfordium	105 Db dubnium	106 Sg seaborgium	107 Bh bohrium	108 Hs hassium	109 Mt meitnerium	110 Ds darmstadtium	111 Rg roentgenium	112 Cn copernicium												
		57 La lanthanum 138.9	58 Ce cerium 140.1	59 Pr praseodymium 140.9	60 Nd neodymium 144.2	61 Pm promethium	62 Sm samarium 150.4	63 Eu europium 152.0	64 Gd gadolinium 157.3	65 Tb terbium 158.9	66 Dy dysprosium 162.5	67 Ho holmium 164.9	68 Er erbium 167.3	69 Tm thulium 168.9	70 Yb ytterbium 173.1	71 Lu lutetium 175.0							
		89 Ac actinium 227.0	90 Th thorium 232.0	91 Pa protactinium 231.0	92 U uranium 238.0	93 Np neptunium	94 Pu plutonium	95 Am americium	96 Cm curium	97 Bk berkelium	98 Cf californium	99 Es einsteinium	100 Fm fermium	101 Md mendelevium	102 No nobelium	103 Lr lawrencium							



Notes

- IUPAC 2009 Standard atomic weights abridged to four significant digits [Table 4 published in *Pure Appl. Chem.* 83, 359-396 (2011); doi:10.1351/PAC-REP-10-09-14]. The uncertainty in the last digit of the standard atomic weight value is listed in parentheses following the value. In the absence of parentheses, the uncertainty is one in that last digit. An interval in square brackets provides the lower and upper bounds of the standard atomic weight for that element. No values are listed for elements with no stable isotopes. See PAC for more details.

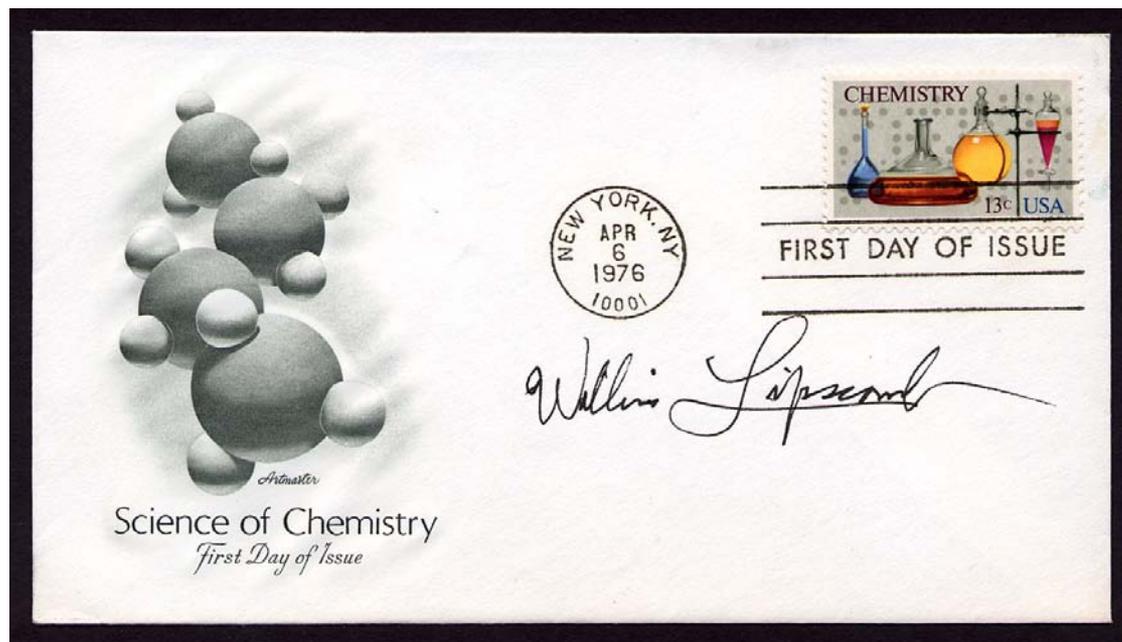
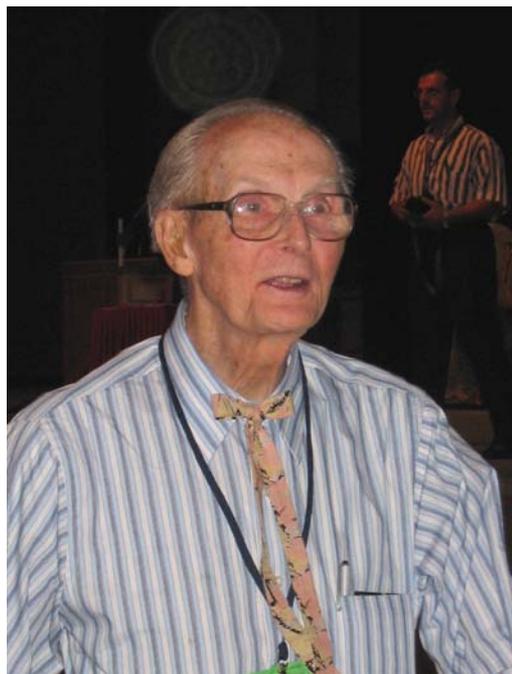
- "Aluminium" and "caesium" are commonly used alternative spellings for "aluminum" and "caesium."

For updates to this table, see iupac.org/reports/periodic_table/. This version is dated 21 January 2011.
Copyright © 2011 IUPAC, the International Union of Pure and Applied Chemistry.



International Year of
CHEMISTRY
2011

*William N. Lipscomb and the
40th IUPAC Congress in Beijing (Aug. '05)*



1976 Nobel Prize in Chemistry “for his studies on the structure of boranes illuminating problems of chemical bonding”

Found on eBay:

a Collection of Chemistry Nobel Laureate Signatures

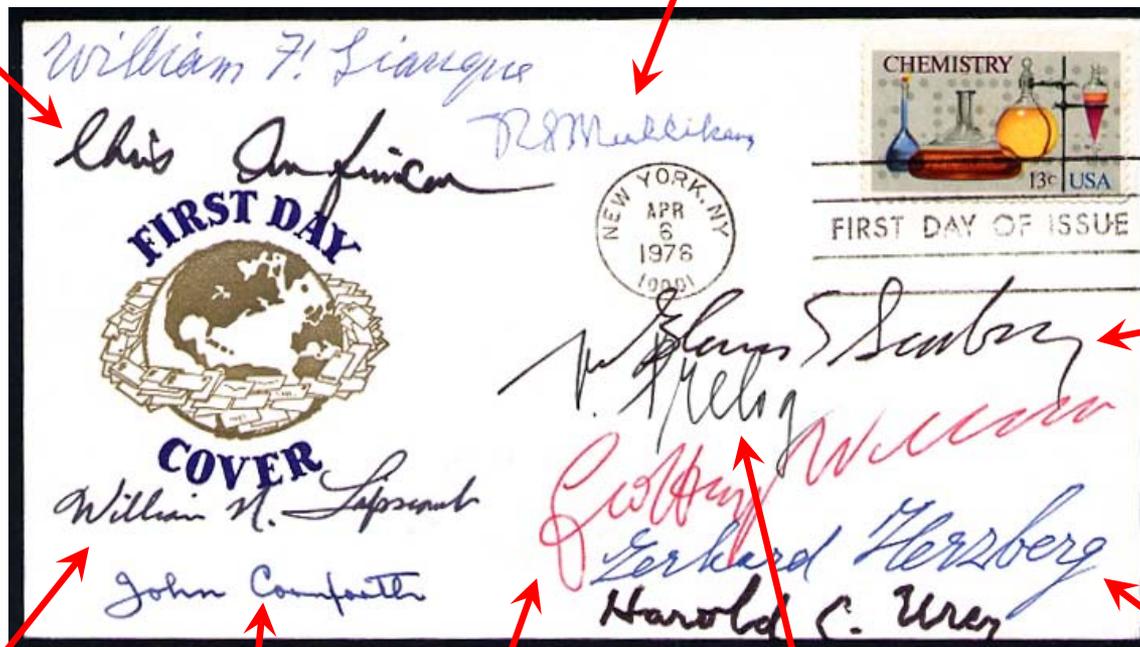


Christian Anfinsen
Chemistry '72
1916-1995

William F. GIAUQUE
Chemistry '49
1895-1982

Robert S. Mulliken
Chemistry '66
1896-1986

Glenn T. Seaborg
Chemistry '51
1912-1999



William N. Lipscomb
Chemistry '76
1919-2011

John Cornforth
Chemistry '75
1917-

Geoffrey Wilkinson
Chemistry '73
1921-1996

Vladimir Prelog
Chemistry '75
1906-1998

Gerhard Herzberg
Chemistry '71
1904-1999

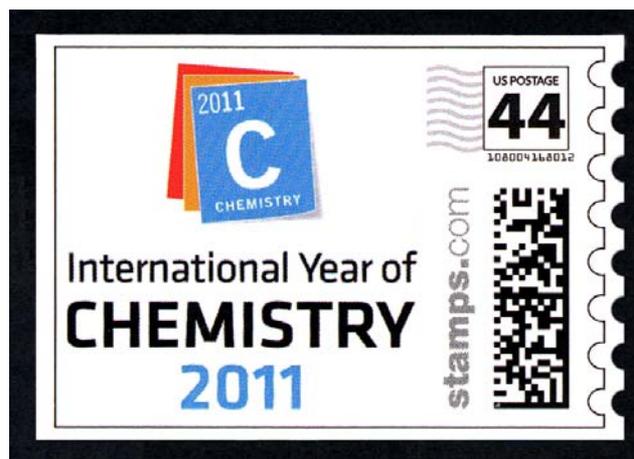
Harold C. Urey
Chemistry '34
1893-1981

IYC Stamps

2011

The International Year of Chemistry (IYC)

www.chemistry2011.org



- IUPAC Project: *Global Stamp Competition*.
- “Chemistry on Stamps” *Symposium and Stamp Exhibition* at the 242nd National ACS meeting in Denver, CO (Fall '11).

First IYC '11 Stamps (4 January 2011): Israel



ubiquitin – protein destructor
Ciechanover & Hershko (Technion)
Chemistry Nobel '04



ribosome – protein constructor
Ada Yonath (WIT)
Chemistry Nobel '09

IYC '11 Stamps: Slovakia



Date of issue:
17 January 2011

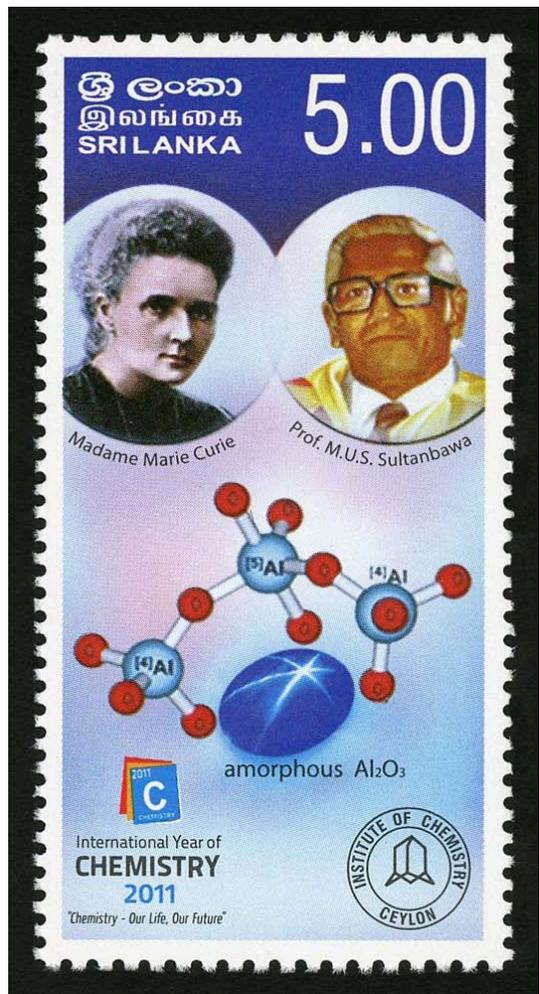
IYC '11 Stamps: Belgium & France



Date of issue:
17 January 2011



Date of issue:
27 January 2011



Date of issue:
30 January 2011

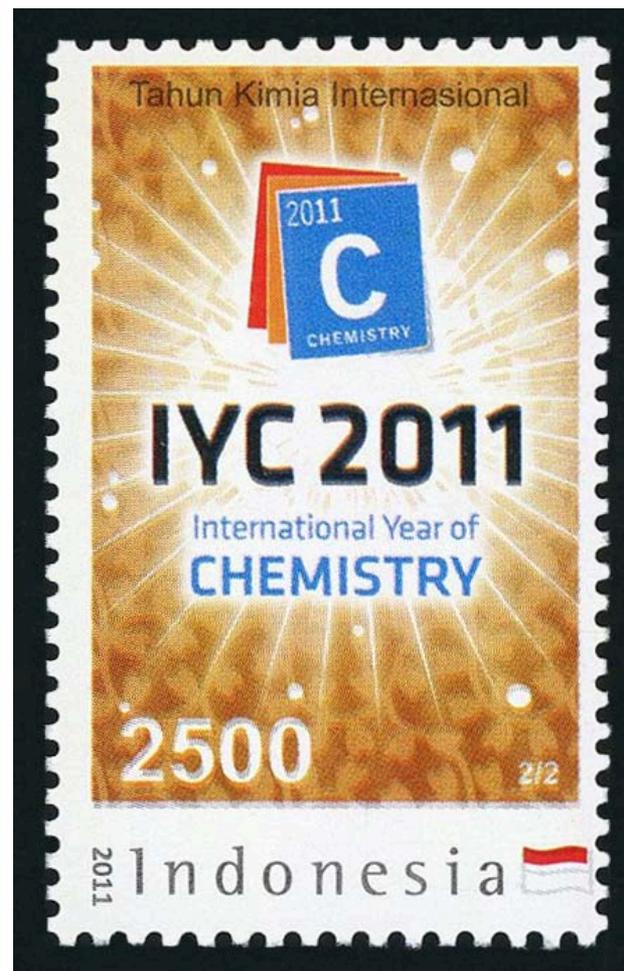
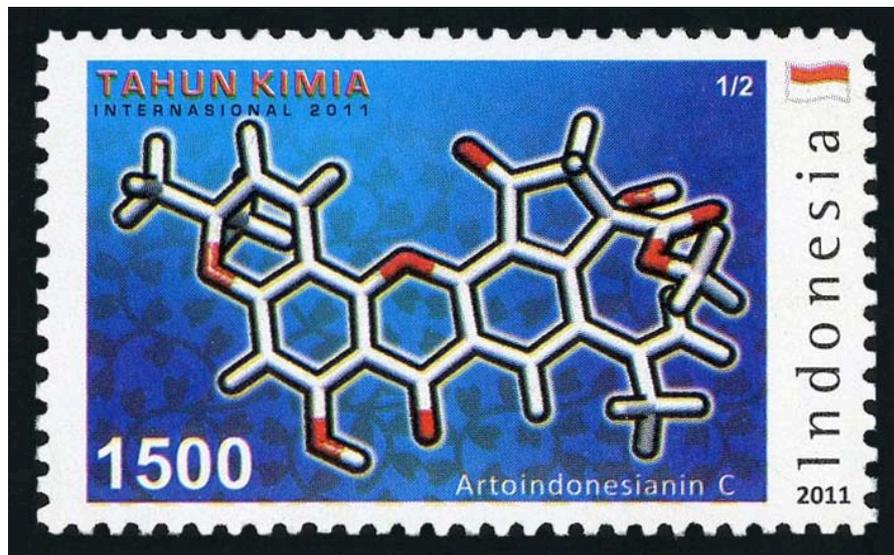
IYC '11 Stamps: Sri Lanka & Spain



Date of issue:
7 February 2011

IYC '11 Stamps: Indonesia

Date of issue:
1 March 2011

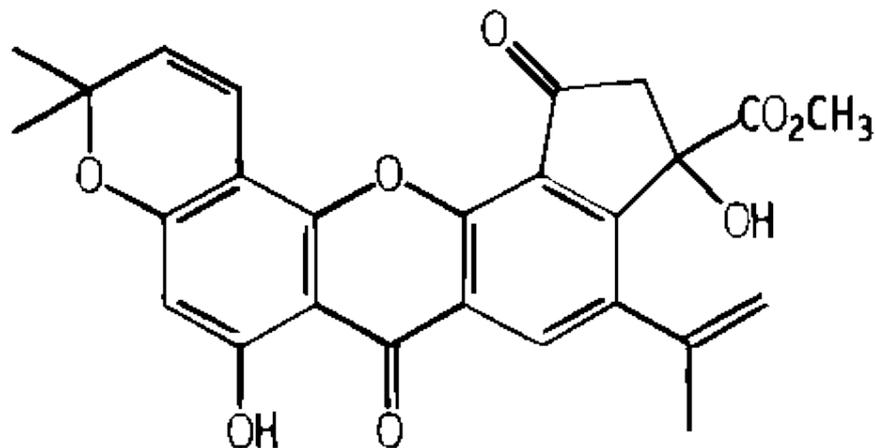
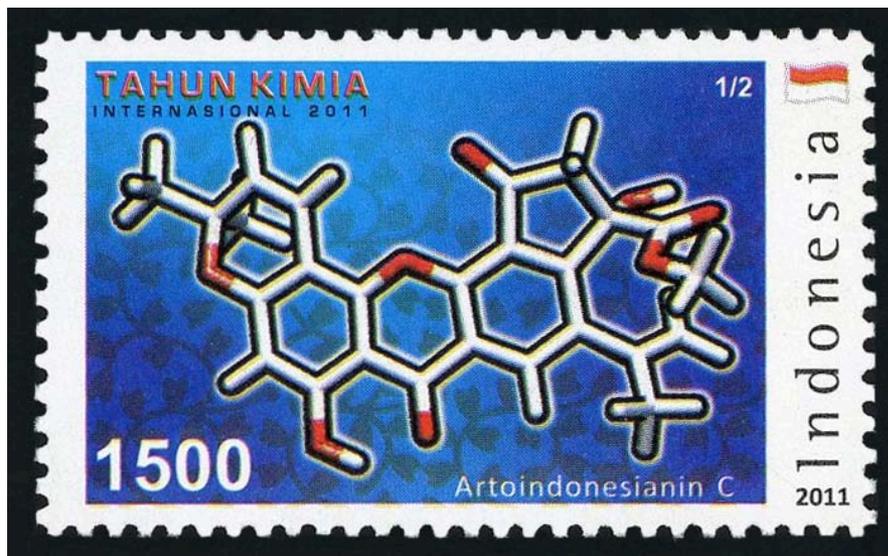


IYC '11 Stamps: Indonesia

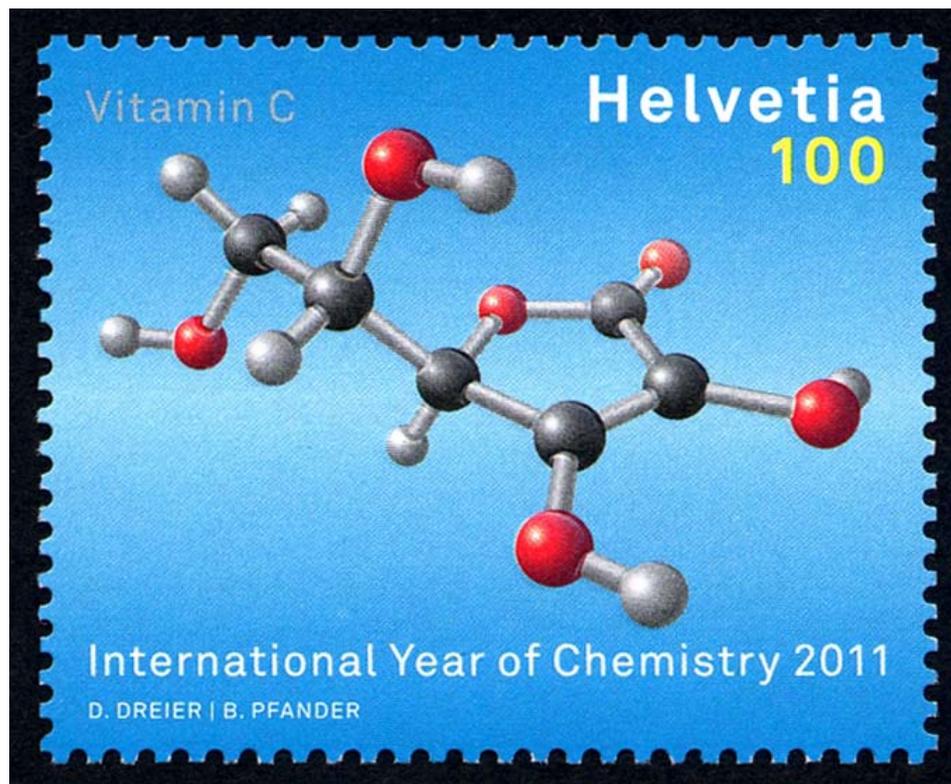
Artoindonesianin C

S.A. Achmad et al.

J. Nat. Prod. **2000**, *63*, 243-244.



IYC '11 Stamps: Switzerland



Date of issue: 3 March 2011



Synthesis of vitamin C: Sir Norman Haworth & Tadeusz Reichstein
Chemistry '37 *Phys or Med '50*

IYC '11 Stamps: Jersey & Bosnia and Herzegovina



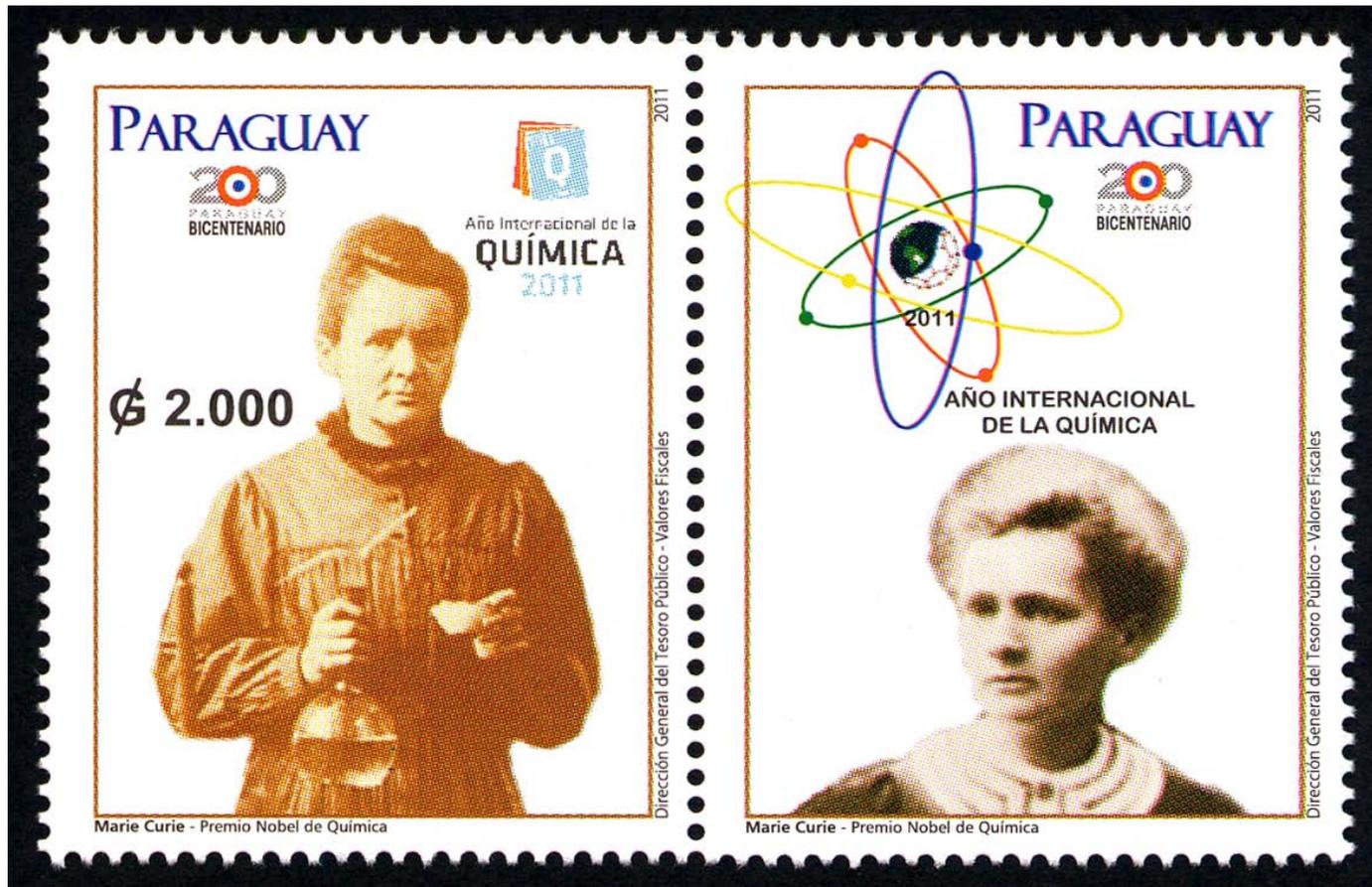
Date of issue: 8 March 2011
(International Women's Day!)

IYC '11 Stamps: Macedonia



Date of issue: 13 April 2011

IYC '11 Stamps: Paraguay



Date of issue: 9 May 2011

The Chemistry and Physics on Stamps Study Unit (**CPOSSU**)

- Non-profit organization that promotes the philatelic study of chemistry, physics, and related fields.
- Started in 1979, currently *ca.* 160 members worldwide.

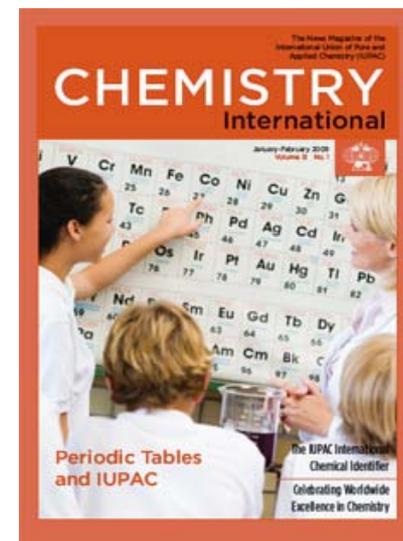
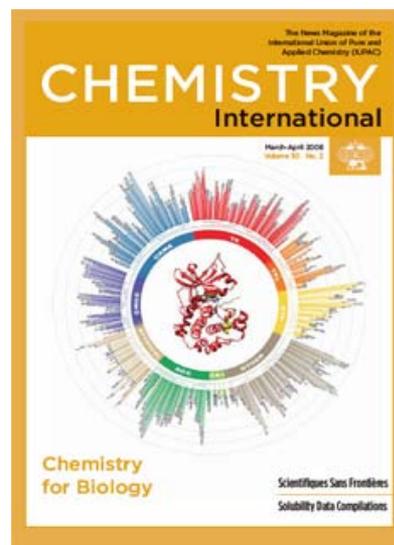
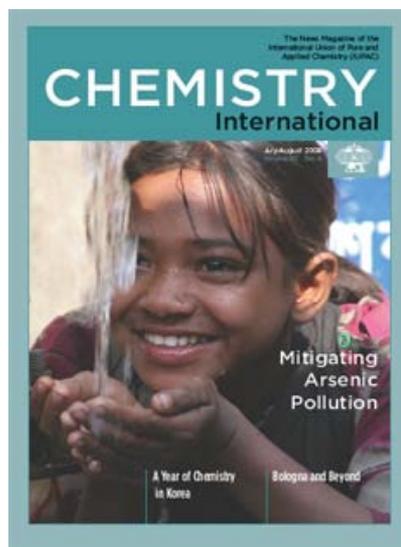
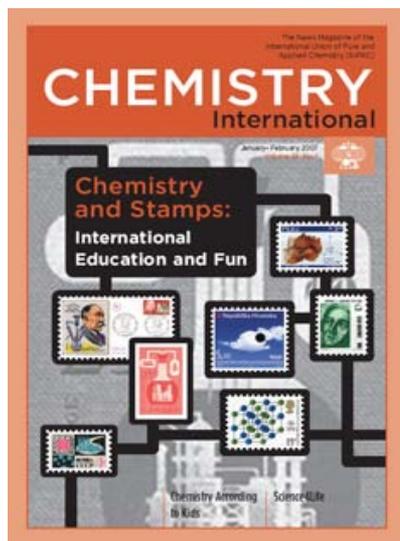
www.cpossu.org

- Publishes *Philatelia Chimica et Physica* (4 issues/year).

Chemistry International

- Bimonthly newsmagazine of IUPAC:

www.iupac.org/publications/ci



Hydrogen to Copernicium:
Postage Stamps as Cultural Icons in the IYC

Daniel Rabinovich

drabinov@uncc.edu

Department of Chemistry
The University of North Carolina at Charlotte

